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- OPTIONAL WIRING
- ..... EXISTING WIRING
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- +----- NEW CONNECTED WIRE

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# Grove Service Manual Amz50xt

**Michal Rosen-Zvi**



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web q1 arrange the bonds in each of the following sets in order of increasing polarity  $\text{c f be f o f o cl s br c p c s b f n o q}_2$  determine based on electronegativity and symmetry if the following molecules are polar or nonpolar  $\text{hcn h c n cf}_4 \text{ c f } 4 \text{ cocl}_2 \text{ c o c l } 2 \text{ nf}_3 \text{ n f } 3 \text{ q}_3$

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as bf 3 trigonal planar ch 4 tetrahedral pf 5 trigonal bipyramidal and sf 6

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