

1.  $\text{FeSO}_4$  \_\_\_\_\_

2.  $\text{FeSO}_4$  \_\_\_\_\_

3.  $\text{FeSO}_4$  \_\_\_\_\_

4.  $\text{FeSO}_4$  \_\_\_\_\_

## EXERCISES

## EXERCISES

### EXERCISES & FORMULAS: Covalent Compounds (2)

a. Write the formulas for the following covalent compounds:

- (1) carbon monoxide \_\_\_\_\_
- (2) carbon dioxide \_\_\_\_\_
- (3) carbon disulfide \_\_\_\_\_
- (4) hydrogen sulfide \_\_\_\_\_
- (5) carbon tetrachloride \_\_\_\_\_
- (6) hydrogen chloride \_\_\_\_\_
- (7) ammonia \_\_\_\_\_
- (8) phosphorus trichloride \_\_\_\_\_

b. Write the names for the following covalent compounds:

- (1)  $\text{CO}$  \_\_\_\_\_
- (2)  $\text{CO}_2$  \_\_\_\_\_
- (3)  $\text{CS}_2$  \_\_\_\_\_
- (4)  $\text{H}_2\text{S}$  \_\_\_\_\_
- (5)  $\text{CCl}_4$  \_\_\_\_\_
- (6)  $\text{HCl}$  \_\_\_\_\_
- (7)  $\text{NH}_3$  \_\_\_\_\_
- (8)  $\text{PCl}_3$  \_\_\_\_\_
- (9)  $\text{PCl}_5$  \_\_\_\_\_

c. Write the covalent names of the following covalent compounds:

- (1)  $\text{CO}$  \_\_\_\_\_
- (2)  $\text{CO}_2$  \_\_\_\_\_
- (3)  $\text{CS}_2$  \_\_\_\_\_
- (4)  $\text{H}_2\text{S}$  \_\_\_\_\_
- (5)  $\text{CCl}_4$  \_\_\_\_\_

# Holt Chemistry Covalent Compounds Review Answers

**F Rizvi**



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