



ANSWER GUIDE
PHYSICAL SCIENCES: GRADE 12
CLASS TEST: ACIDS & BASES

Written on: 25.06.2020

Total Marks: 65

1	1.1	B ✓✓	2
	1.2	C ✓✓	2
2	Hydrochloric acid is a strong acid and sodium hydroxide is a strong base, ✓ therefore the equivalence point will be reached at a pH close to 7, ✓ which is in the colour change range of bromothymol blue.		2
3	3.1	An acid is a proton donor. ✓	1
	3.2	A concentrated acid is an acid solution with a high proportion of acid (solute) to water (solvent), ✓✓ regardless of the degree of ionisation or dissociation of the acid in the solution. A strong acid is an acid that dissociates or ionises almost completely when in solution, regardless of its concentration. ✓✓	4
4	4.1	LiOH ✓	5
	4.2	H ₂ PO ₄ ✓	
	4.3	HNO ₃ ✓	
	4.4	HCOOH ✓	
	4.5	NaHCO ₃ ✓	
5	$\text{H}_2\text{SO}_4 + 2\text{H}_2\text{O} \checkmark \rightarrow 2\text{H}_3\text{O}^+ \checkmark + \text{SO}_4^{2-} \checkmark$ OR: $\text{H}_2\text{SO}_4 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{HSO}_4^-$		3
6	6.1	A standard solution is a solution of known concentration. ✓✓	2
	6.2	$c = \frac{m}{MV} \quad \therefore m = cMV \checkmark$ $= (0,15)(106)(0,25) \checkmark$ $= \underline{3,975 \text{ g Na}_2\text{CO}_3} \checkmark$	3
	6.3	$\frac{n_a}{n_b} = \frac{c_a V_a}{c_b V_b} \quad \therefore c_a = \frac{n_a c_b V_b}{n_b V_a} \checkmark$ $= \frac{(1)(0,15)(25)}{(1)(27,5)} \checkmark$ $= \underline{1,38 \times 10^{-1} \text{ mol.dm}^{-3}} \checkmark$	4
7	$\text{pH} = -\log[\text{H}^+] \checkmark$ $= -\log[0,2] \checkmark$ $= \underline{0,69} \checkmark$		3
	and: HCl dissociates completely $\therefore 0,2 \text{ mol.dm}^{-3} \text{ H}^+$ ions		

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Table of Contents Holt Chemistry Acid And Bases Test Answers

1. Understanding the eBook Holt Chemistry Acid And Bases Test Answers
 - The Rise of Digital Reading Holt Chemistry Acid And Bases Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Acid And Bases Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Acid And Bases Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Acid And Bases Test Answers

- Personalized Recommendations
 - Holt Chemistry Acid And Bases Test Answers User Reviews and Ratings
 - Holt Chemistry Acid And Bases Test Answers and Bestseller Lists
5. Accessing Holt Chemistry Acid And Bases Test Answers Free and Paid eBooks
 - Holt Chemistry Acid And Bases Test Answers Public Domain eBooks
 - Holt Chemistry Acid And Bases Test Answers eBook Subscription Services
 - Holt Chemistry Acid And Bases Test Answers Budget-Friendly Options
 6. Navigating Holt Chemistry Acid And Bases Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Chemistry Acid And Bases Test Answers Compatibility with Devices
 - Holt Chemistry Acid And Bases Test Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Chemistry Acid And Bases Test Answers
 - Highlighting and Note-Taking Holt Chemistry Acid And Bases Test Answers
 - Interactive Elements Holt Chemistry Acid And Bases Test Answers
 8. Staying Engaged with Holt Chemistry Acid And Bases Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Acid And Bases Test Answers
 9. Balancing eBooks and Physical Books Holt Chemistry Acid And Bases Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Chemistry Acid And Bases Test Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Holt Chemistry Acid And Bases Test Answers
 - Setting Reading Goals Holt Chemistry Acid And Bases Test Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Holt Chemistry Acid And Bases Test Answers

- Fact-Checking eBook Content of Holt Chemistry Acid And Bases Test Answers
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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