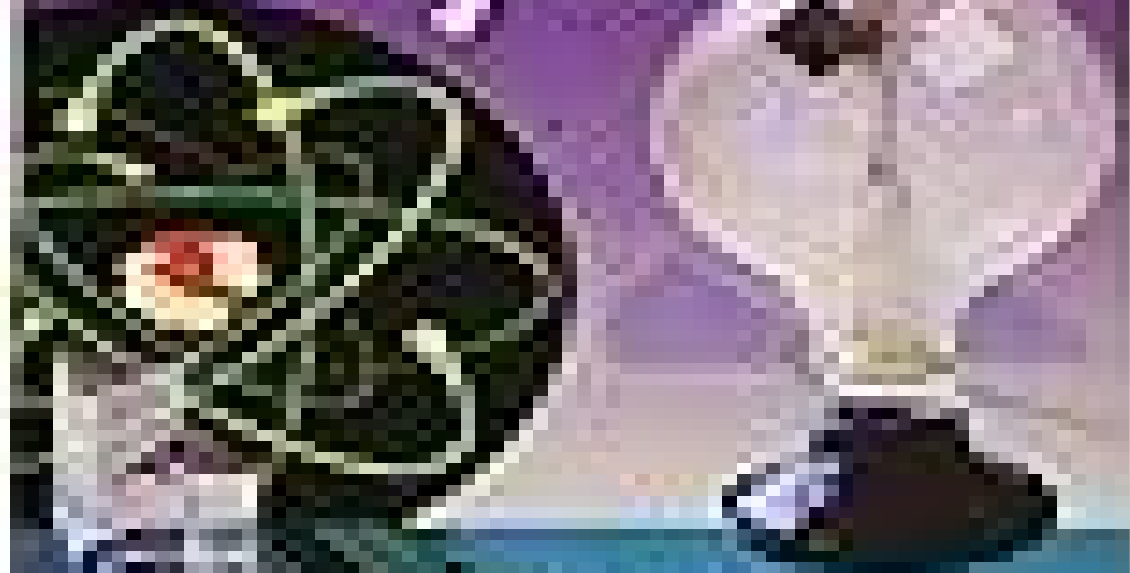


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web $q = 140 \text{ g} \times 4.184 \text{ J/g} \times 16^\circ\text{C}$ $q = 9372 \text{ J}$ $35.5 \text{ g} \times 1 \text{ mole KCN} / 0.5453 \text{ moles}$ $65.1 \text{ g} / 0.5453 \text{ mole moles}$ 5 when a 15 g sample of $SnCl_2$ dissolves in 80 g of water in a calorimeter the temperature drops from 42°C to 15°C calculate the ΔH for the process
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web thermochemistry answers and explanations 1 a thermodynamic system has two means of exchanging energy with its surroundings heat flow and pressure volume work heat flow q is the transfer of thermal energy due to temperature difference work w is macroscopic energy transfer between the system and its surroundings that occurs

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web jan 10 2021 $\Delta H = 274 \text{ kJ/mol}$ and $\Delta S = 298 \text{ J/K}$ sulfur trioxide SO_3 is produced in large quantities in the industrial synthesis of sulfuric acid sulfur dioxide is converted to sulfur trioxide by reaction with oxygen gas write a balanced chemical equation for the reaction of SO_2 with $O_2(g)$ and determine its ΔG