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chemical equation $2\text{Fe}_2\text{O}_3 + \text{C} \rightarrow \text{Fe} + 3\text{CO}_2$ on towards the next step now step 3 once you are done with this check out if the number
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 $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$ 2 $\text{C}_6\text{H}_{10}\text{O}_2 + \text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$ 3 $\text{GaBr} + \text{NaBr} \rightarrow \text{SnO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{S}_3 + \text{SO}_3$ 4 $\text{NF}_3 + \text{SnF}_2 \rightarrow \text{N}_2\text{O}_3$ using the equation
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