

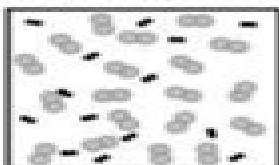
CHAPTER 18 HOW CHEMICALS MIX

Multiple Choice

Identify the choice that best completes the statement or answers the question.

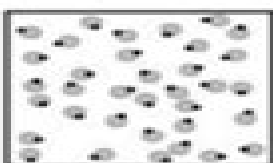
- _____ 1. A combination of two or more substances in which they no longer retain their chemical properties is called a(n) _____.
a. heterogeneous mixture
b. compound
c. suspension
d. mixture
e. periodic trend

- _____ 2. The following image represents which kind of matter?



- a. a mixture
b. an element
c. a compound
d. none of the above
e. all of the above

- _____ 3. The following image represents which kind of matter?



- a. an element
b. a compound
c. a mixture
d. none of the above
e. all of the above

- _____ 4. Which of the following is a mixture?

- a. salt
b. iron
c. gold
d. helium
e. air

- _____ 5. How does a suspension differ from a solution?

- a. A suspension is a heterogeneous mixture consisting of different phases whereas a solution is a homogeneous mixture consisting of a single phase.
b. The difference between a suspension and a solution can only be determined by chemical means.
c. Although a solution and suspension are both homogeneous mixtures, only the components of a suspension will separate by spinning the mixture in a centrifuge.
d. A suspension is a heterogeneous mixture whose components can be separated by simple filtration. A solution is a homogeneous mixture which cannot be separated by simple

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