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web charles law when the temperature of a gas is increased at constant pressure its volume increases when the temperature of a gas is decreased at constant pressure its volume decreases boyle s law when the pressure of a gas is at a constant temperature is increased the volume decreases

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web interactive textbook answer key 70 physical science physical science answer key continued 3 the particles of a liquid can move past one another but the particles of a solid stay in fixed positions 4 the particles of a gas can move far away from one another but the particles of a liquid stay close to one another 5 surface tension 6

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web learn test match q chat created by nifemi adetunji terms in this set 91 compressibility a measure of how much the volume of matter decreases under pressure because of the space between particles in a gas why are gases easily compressed straight line paths until they collide with other particles or the walls of their container

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web 1 look at figure 14.1 on page 413 how does an automobile air bag protect the crash dummy from being broken as a result of impact the gases used to inflate the airbag are able to absorb a considerable amount of energy when they are compressed 2 what theory explains the behavior of gases kinetic theory 3

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web class states of matter behavior of gases before you read after you read this section you should be able to answer these questions what affects how a gas behaves what are the gas laws what affects the behavior of a gas gases behave differently than solids or liquids gas particles have a large amount of space between them

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web the volume of a gas is the same as the volume of its container a pressure of 101.3 kPa is equal to the air inside a fully pumped basketball has a higher pressure than the air outside because there is a higher number of gas particles per unit volume does the air inside a fully pumped basketball have a higher or lower pressure than the air

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web the behavior of gases directions answer each question on the lines provided you must include the terms below in your answer boyle's law charles's law kinetic molecular theory pressure 1 temperature pressure and volume affect the behavior of gases which variable is held constant in the relationship described by boyle's law 2

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web the amount of space that matter fills measured in cubic centimeters cm<sup>3</sup> milliliters ml liters l gas particles move and fill the space available this measure of gas is the same as the measure of its container a measure of the average energy of random motion of the particles of a substance

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