

Unit 4, Lab 2: Boyle's Law

Name: _____

Date: _____

Boyle's Law: In a closed system, the relationship between the volume and pressure of a gas is:

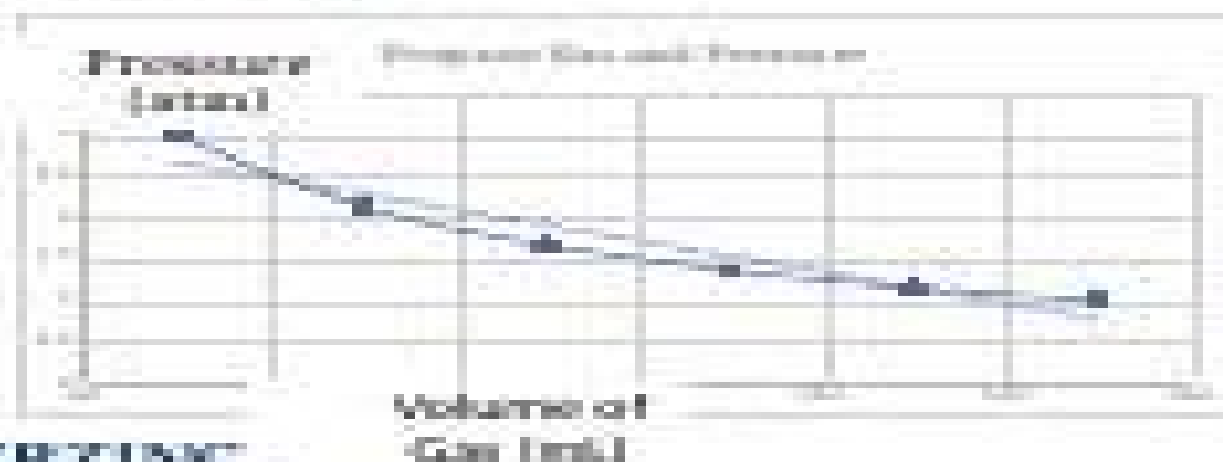
Lab Objectives:

1. Demonstrate the relationship between pressure and volume of a gas.

Pressure (atm)	Volume (mL)
1.00 atm	1.00 mL
0.80 atm	1.25 mL
0.60 atm	1.67 mL
0.40 atm	2.50 mL
0.20 atm	5.00 mL
0.10 atm	10.00 mL

Data Analysis:

2. Plot the pressure-volume data on a graph. The x-axis is Volume (mL) and the y-axis is Pressure (atm). The graph should show a hyperbolic curve, indicating an inverse relationship between pressure and volume.



Late Night Lab Chemistry Answers

Tobias Bleicker



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