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Bill Tillery



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the microscale method This approach to chemistry gives students a lab experience as good as or better than the traditional methods but uses about 1/100th of the chemicals The experiments are much safer and disposal much easier The chemical solutions are pre mixed and in dropping bottles that give constant drop size This eliminates the need to mix solutions greatly reduces spills and reduces set up time to a few minutes

Introduction Lab Melting Points Super Cooling 1 Empirical Formula 2 Analysis of Hydrates 3 Molar Mass by Titration 4 Freezing Point Depression 5 Gas Laws Boyle's Law 6 Gas Laws Charles's Law 7 Molar Volume of a Gas 8 A Standard Acid and a Standardized Base 9 A Microscale Titration 10 A Weak Acid Strong Base Titration 11 Oxidation Reduction 12 Mole Ratios 13 Double Replacement Reactions 14 Solubility Product Constant 15 pH and pH Indicators 16 Reaction Rates The Effect of Concentration 17 Reaction Rates The Effects of Temperature and Particle Size 18 Radioactive Decay 19 Enthalpy of Fusion of Ice 20 Decomposition of H₂O and NaCl 21 Properties of Cations and Anions 22 Synthesis of a Coordination Compound 23 Synthesis and Analysis of Aspirin 24 Gravimetric Analysis 25 Colorimetry 26 Paper Chromatography 27 A Buffer Solution 28 Electrical Conductivity of Several Solutions 29 Electrochemistry Galvanic Cells

Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2011-06-21

This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

Lab Manual for Chemistry: Atoms First John W Sibert, Gregg Dieckmann, 2014-01-13 Laboratory Manual to Accompany Chemistry Atoms First by Gregg Dieckmann and John Sibert from the University of Texas at Dallas This laboratory manual presents a lab curriculum that is organised around an atoms first approach to general chemistry The philosophy behind this manual is to 1 provide engaging experiments that tap into student curiosity 2 emphasize topics that students find challenging in the general chemistry lecture course and 3 create a laboratory environment that encourages students to solve puzzles or play with course content and not just follow recipes The laboratory manual represents a terrific opportunity to get students turned on to science while creating an environment that connects the relevance of the experiments to a greater understanding of their world This manual has been written to provide instructors with tools that engage students while providing important connections to the material covered in an atoms first lecture course

General Chemistry Laboratory Manual Petra Van-Koppen, 2003 Presents a lab manual for the two semester General Chemistry course This book contains experiments that cover the commonly assigned experiments found in a typical two semester course

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