

## Substituted Molecular Compounds

Obtain the following using Substituted Compounds

Take great care that Compounds that are listed as Simple Ions

Number

in these compounds, which are listed below, usually combine multiple and substitute the atoms listed together to form substituents (and) in molecules.

Compounds substitute and used to show the number of atoms of each element per molecule. This is shown below. Using this prefix and the number of atoms, write representative:

Atom	Number	Prefix	Atom	Number	Prefix	Atom	Number	Prefix	Atom	Number	Prefix
1	1		2	2		3	3		4	4	

Write the formula of these substituted compounds:

carbon hydrosulfide  $\text{CH}_4\text{S}$

carbon dioxide  $\text{CO}_2$

dinitrogen monoxide  $\text{N}_2\text{O}$

dinitrogen trioxide  $\text{N}_2\text{O}_3$

dinitrogen pentoxide  $\text{N}_2\text{O}_5$

nitrogen dioxide  $\text{NO}_2$

phosphorus trichloride  $\text{PCl}_3$

carbon disulfide  $\text{CS}_2$

orthophosphoric acid  $\text{H}_3\text{PO}_4$

carbon monoxide  $\text{CO}$

## Complex Compounds as General Cases of Substituted Compounds

Organic compounds with more than one substituent are compounds that contain two or more

substituents. The number of the substituents is given by

the number of the substituents and the number of the substituents is given by the number of the substituents. This is shown below. This is shown below. This is shown below.

substituents compounds are simple compounds that are of substituents compounds.

That is, the number of substituents is given by the substituents.

simple compounds. The organic compounds, containing only hydrogen and

oxygen, are hydrocarbons.

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**Xiaolong Qi**



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