

### Final Exam Study Guide (General Chemistry I)

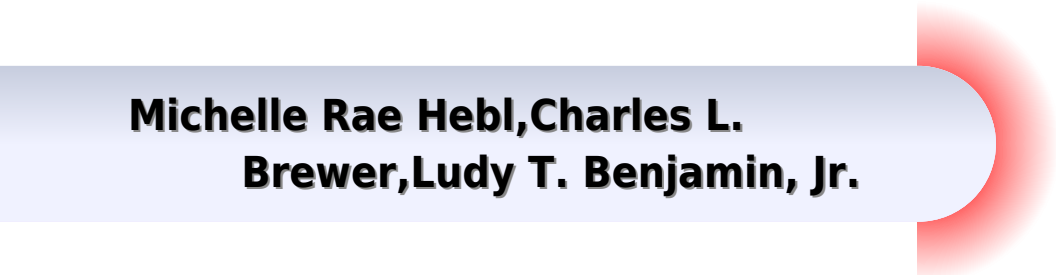
- Chemistry - the study of matter and the changes that matter undergoes
- Matter - anything that has mass and takes up space
  - Three states of matter: solid, liquid, and gas (plasma is a gas)

	Proximity of Particles	Energetics	Size/Shape
Solid	Closely packed and in direct contact with each other	Have very little energy of motion (slow)	Fixed shape and fixed volume
Liquid	In indirect contact with each other; usually about a particles distance apart (bounce off each other)	Have an intermediate amount energy of motion (medium)	Does not have a fixed shape; has a fixed volume
Gas	Particles are very far apart	Have a large amount of energy of motion (fast)	No fixed shape or volume

- **PARTICLES ARE ALWAYS MOVING!**
- "Can you undo it?" (Page 12)
  - Physical Changes - changes physical appearance but not its composition
    - Ex. Melting, evaporation, freezing, etc.
  - Chemical Changes - changes in chemical composition of matter (chemical reaction)
    - Ex. Baking soda and vinegar, burning, mixing, etc.
  - Physical Properties - characteristics that can be described by observing the matter; no change in composition is required to detect a physical property
    - Ex. Color, odor, density, etc.
  - Chemical Properties - a change in composition is required to detect a chemical property
    - Ex. Flammability
- Accuracy Versus Precision:
  - In every measurement, there is some degree of uncertainty.
  - **NUMBERS WITHOUT UNITS ARE MEANINGLESS IN MEASUREMENT!**
  - Accuracy - the agreement of a measurement with a known or accepted value (correctness)
  - Precision - the agreement of a measurement with other measurements in a given set of measurements (reproducibility/repeatability)
  - You can have accuracy without precision, and precision without accuracy.
  - **WE ARE ONLY AS CERTAIN AS OUR LEAST CERTAIN NUMBER!**
    - 254 can mean 253.5 to 254.4
    - 254.0 can mean 253.95 to 254.04
    - 254.00 can mean 253.950 to 254.004
    - 254.000 can mean 253.9500 to 254.0004
- Density - the mathematical relationship of mass to volume

# Grade 12 Chemistry Final Exam Study Guide

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## **Grade 12 Chemistry Final Exam Study Guide:**

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*Handbook for Teaching Introductory Psychology* Michelle Rae Hebl, Charles L. Brewer, Ludy T. Benjamin, Jr., 2001-08-01 Like its predecessors Volume III of the Handbook for Teaching Introductory Psychology provides introductory psychology instructors with teaching ideas and activities that can immediately be put into practice in the classroom It contains an organized collection of articles from Teaching of Psychology TOP the official journal of the Society for the Teaching of Psychology Division 2 of the American Psychological Association Volume III contains 89 articles from TOP that have not been included in other volumes Another distinction between this volume and its predecessors is its emphasis on testing and assessment The book is divided into two sections Section One Issues and Approaches in Teaching Introductory Psychology contains 52 articles on critical issues such as how to approach the course understanding students interests perceptions and motives students existing knowledge of psychology including their misconceptions a comparison of introductory textbooks and tips on how to evaluate them test questions and student factors affecting exam performance an overview of different forms of feedback giving extra credit and how to deal with academic dishonesty Section Two consists of 37 articles that present demonstrations class and laboratory projects and other techniques to enhance teaching and learning in both the introductory as well as advanced courses in the discipline This section is organized so as to parallel the order of topics found in most introductory psychology textbooks Intended for academicians who teach the introductory psychology course and or oversee grad assistants who teach the course all royalties of the book go directly to the Society for the Teaching of Psychology to promote its activities to further improve the teaching of psychology

*Resources in Education*, 1998

*X-kit Literature Series: FET Maru* Colleen Traviss-Lea, 2005

*Carolina Science and Math* Carolina Biological Supply Company, 2003

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## **Table of Contents Grade 12 Chemistry Final Exam Study Guide**

1. Understanding the eBook Grade 12 Chemistry Final Exam Study Guide
  - The Rise of Digital Reading Grade 12 Chemistry Final Exam Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Grade 12 Chemistry Final Exam Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grade 12 Chemistry Final Exam Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 12 Chemistry Final Exam Study Guide
  - Personalized Recommendations
  - Grade 12 Chemistry Final Exam Study Guide User Reviews and Ratings
  - Grade 12 Chemistry Final Exam Study Guide and Bestseller Lists
5. Accessing Grade 12 Chemistry Final Exam Study Guide Free and Paid eBooks
  - Grade 12 Chemistry Final Exam Study Guide Public Domain eBooks
  - Grade 12 Chemistry Final Exam Study Guide eBook Subscription Services
  - Grade 12 Chemistry Final Exam Study Guide Budget-Friendly Options
6. Navigating Grade 12 Chemistry Final Exam Study Guide eBook Formats
  - ePub, PDF, MOBI, and More
  - Grade 12 Chemistry Final Exam Study Guide Compatibility with Devices
  - Grade 12 Chemistry Final Exam Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Grade 12 Chemistry Final Exam Study Guide
  - Highlighting and Note-Taking Grade 12 Chemistry Final Exam Study Guide
  - Interactive Elements Grade 12 Chemistry Final Exam Study Guide
8. Staying Engaged with Grade 12 Chemistry Final Exam Study Guide

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Grade 12 Chemistry Final Exam Study Guide
- 9. Balancing eBooks and Physical Books Grade 12 Chemistry Final Exam Study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Grade 12 Chemistry Final Exam Study Guide
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 12 Chemistry Final Exam Study Guide
  - Setting Reading Goals Grade 12 Chemistry Final Exam Study Guide
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 12 Chemistry Final Exam Study Guide
  - Fact-Checking eBook Content of Grade 12 Chemistry Final Exam Study Guide
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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