

Experiment: Preparation of esters

Answer the following questions related to the experiment.

1	What is the function of the conc. H_2SO_4 ?	(2)
2	Why were the reactants of each test tube heated in a water bath and not directly over a flame?	(2)
3	Is the organic product soluble in water?	(1)
4	For each of the following structures below, identify the acid and alcohol used to prepare these esters. Also give the name of the ester.	
4.1	$ \begin{array}{ccccccc} & H & O & & & CH_3 & \\ & & & & & & \\ H & - C - & C - O - CH_2 - CH_2 - & CH - CH_3 \\ & & & & & & \\ & H & & & & CH_3 & \\ & & & & & \text{Benzene} & \end{array} $ Name of the acid used: Name of the alcohol used: Name of the Ester:	(6)
4.2	$ \begin{array}{ccccccc} & O & & & CH_3 & & \\ & & & & & & \\ H & - C - & O - C - & CH_3 \\ & & & & & & \\ & & & H & & & \end{array} $ Name of the acid used: Name of the alcohol used: Name of the Ester:	(6)
4.3	$ \begin{array}{ccccccc} & H & O & & & & \\ & & & & & & \\ H & - C - & C - O - CH_2 - (CH_2)_8 - CH_3 \\ & & & & & & \\ & H & & & & & \\ & & & & & \text{Orange} & \end{array} $ Name of the acid used: Name of the alcohol used:	

Grade12 Experiment Esters

Daniel F McAuley

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Grade12 Experiment Esters

1. Understanding the eBook Grade12 Experiment Esters
 - The Rise of Digital Reading Grade12 Experiment Esters
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade12 Experiment Esters
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade12 Experiment Esters
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade12 Experiment Esters
 - Personalized Recommendations
 - Grade12 Experiment Esters User Reviews and Ratings
 - Grade12 Experiment Esters and Bestseller Lists
5. Accessing Grade12 Experiment Esters Free and Paid eBooks
 - Grade12 Experiment Esters Public Domain eBooks

- Grade12 Experiment Esters eBook Subscription Services
- Grade12 Experiment Esters Budget-Friendly Options
- 6. Navigating Grade12 Experiment Esters eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade12 Experiment Esters Compatibility with Devices
 - Grade12 Experiment Esters Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade12 Experiment Esters
 - Highlighting and Note-Taking Grade12 Experiment Esters
 - Interactive Elements Grade12 Experiment Esters
- 8. Staying Engaged with Grade12 Experiment Esters
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade12 Experiment Esters
- 9. Balancing eBooks and Physical Books Grade12 Experiment Esters
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade12 Experiment Esters
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade12 Experiment Esters
 - Setting Reading Goals Grade12 Experiment Esters
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade12 Experiment Esters
 - Fact-Checking eBook Content of Grade12 Experiment Esters
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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