

Chapter 8 Chemical Reactions Chapter Review Answers

1. The evolution of energy, the production of a gas, the formation of a precipitate, and a color change.
2. A correctly written chemical equation must represent the known facts, contain the correct formulas for the reactants and products, and satisfy the law of conservation of mass.
3. a. It is a number that appears in front of a formula in a chemical equation.
b. The number of atoms of each type in the formula is multiplied by the coefficient.
4. methane + oxygen $\rightarrow$ carbon dioxide + water;
 $\text{CH}_4(\text{g}) + \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g})$;
 $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{g})$
5. A chemical equation provides the number of molecules, moles, and atoms of each element or compound composing the reactants and products as well as the mole ratios of the reactants to each other and to the products.
6. Word and formula equations have only qualitative meaning; they do not express the relative quantities of a reaction's reactants or products.
7. a. a solution in water
b. a substance that accelerates a chemical reaction but can be recovered unchanged
c. a chemical reaction in which the products re-form the original reactants
8. a. KOH d. CCl_4
b. $\text{Ca}(\text{NO}_3)_2$ e. MgBr_2
c. Na_2CO_3
9. (1) Balance different types of atoms one at a time; (2) balance types of atoms that appear only once on each side of the equation first; (3) balance as single units any polyatomic ions that appear on both sides of the equation; and (4) balance H atoms and O atoms last.
10. a. 6N
b. 4H, 2O
c. 4H, 4N, 12O
d. 2Ca, 4O, 4H
e. 3Ba, 6Cl, 18O
f. 5Fe, 10N, 30O
g. 12Mg, 8P, 32O
h. 4N, 16H, 2S, 8O
i. 12Al, 18Se, 72O
j. 12C, 32H
11. a. $2\text{ZnS}(\text{s}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{ZnO}(\text{s}) + 2\text{SO}_2(\text{g})$
b. $2\text{HCl}(\text{aq}) + \text{Ba}(\text{OH})_2(\text{aq}) \rightarrow \text{BaCl}_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$
c. $2\text{HNO}_3(\text{aq}) + \text{Ca}(\text{OH})_2(\text{aq}) \rightarrow \text{Ca}(\text{NO}_3)_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$
12. a. Solid zinc sulfide reacts with oxygen gas to form solid zinc oxide and sulfur dioxide gas.
b. When solid calcium hydride is added to water, aqueous calcium hydroxide and hydrogen gas are formed.
c. Aqueous silver nitrate mixed with aqueous potassium iodide produces a precipitate of silver iodide and aqueous potassium nitrate.
13. a. $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
b. $2\text{Al} + \text{Fe}_2\text{O}_3 \rightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$
c. $\text{Pb}(\text{CH}_3\text{COO})_2 + \text{H}_2\text{S} \rightarrow \text{PbS} + 2\text{CH}_3\text{COOH}$
14. a. LiO_2 is an incorrect formula; $4\text{Li} + \text{O}_2 \rightarrow 2\text{Li}_2\text{O}$
b. H_2Cl_2 is an incorrect formula; $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
c. MgO_2 is an incorrect formula, and the equation as written is not balanced; $\text{MgCO}_3 \rightarrow \text{MgO} + \text{CO}_2$
d. I is an incorrect formula for iodine, and the equation is not balanced; $2\text{NaI} + \text{Cl}_2 \rightarrow 2\text{NaCl} + \text{I}_2$
15. a. $4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$
b. $\text{P}_4\text{O}_{10} + 6\text{H}_2\text{O} \rightarrow 4\text{H}_3\text{PO}_4$
c. $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$
16. $\text{CH}_4(\text{g}) + 4\text{Cl}_2(\text{g}) \rightarrow \text{CCl}_4(\text{l}) + 4\text{HCl}(\text{g})$
17. a. $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$
b. $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
c. $2\text{Li} + \text{Cl}_2 \rightarrow 2\text{LiCl}$
d. $\text{Ca} + \text{I}_2 \rightarrow \text{CaI}_2$
18. (1) In a synthesis reaction, two or more substances combine to form a new substance: $\text{A} + \text{X} \rightarrow \text{AX}$.
(2) In a decomposition reaction, a single compound undergoes a reaction that produces two or more simpler substances: $\text{AX} \rightarrow \text{A} + \text{X}$.
(3) In a single-displacement reaction, one element replaces another element in a compound:
 $\text{A} + \text{BX} \rightarrow \text{AX} + \text{B}$ or $\text{Y} + \text{BX} \rightarrow \text{BY} + \text{X}$

Health Chemistry 8 Chapter Review Answers

KJ Lindholm-Leary



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Rudolph's Pediatrics Self-Assessment and Board Review Michael D. Cabana, 2013-12-11 Complemented by Rudolph's pediatrics editor in chief Colin D Rudolph editors Abraham M Rudolph et al 22nd ed c2011 **Chemistry** Karen Timberlake, 2003 Chemistry An Introduction to General Organic and Biological Chemistry now in its eighth edition makes chemistry exciting by showing why important concepts are relevant to the lives and future careers of readers The new design digital images photos Career Focus features and macro to micro art enhance the new edition while it retains the many features that have made this book so successful The writing as always is exceptionally friendly Each section contains sample problems that develop readers critical thinking skills This edition also contains more conceptual problems than ever before and has been redesigned to accommodate new styles of learning and teaching with a wide variety of pedagogical tools Health and Environmental Notes throughout the book highlight topics that are relevant to readers lives and are ideal for classroom discussion Explore Your World activities in each chapter make chemistry exciting relevant and non threatening

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Modern Hydrotherapy for the Massage Therapist Marybetts Sinclair, 2007-11-01 This text offers a modern approach to hydrotherapy the use of water ice steam and hot and cold temperatures to improve or maintain health as performed by massage therapists Authored by an experienced massage therapist this book presents clear uncomplicated explanations of how hydrotherapy affects the body and then demonstrates a wide variety of hydrotherapy treatments The book suggests how massage therapists may use hydrotherapy treatments before and during massage sessions or give these treatments to clients to do between sessions for faster and better results The author presents real life examples and case studies obtained through interviews with massage therapists athletic trainers physical therapists naturopathic doctors aquatic therapists and medical doctors treating patients in a medical hydrotherapy setting **Phlebotomy Exam Review**

Ruth E. McCall, 2023-08-15 Phlebotomy Exam Review Eighth Edition provides a comprehensive review of current phlebotomy theory and offers an ideal way to study for phlebotomy licensing or national certification exams. It also makes for an excellent study tool for students taking formal phlebotomy training programs. By answering the questions in this review, a student can test their knowledge and application of current phlebotomy theory. Theory questions address recent federal safety standards, Clinical and Laboratory Standards Institute (CLSI) guidelines, and the National Accrediting Agency for Clinical Laboratory Sciences (NAACLS) phlebotomist competencies when applicable. Questions are standard multiple choice like those used on national exams with choices that often test your critical thinking abilities. 2024 380 pages

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Your Total ASVAB Solution Wallie Walker-Hammond, 2011-09-09 Attention Military Applicants REA has Your Total ASVAB Solution Your Total ASVAB Solution Helps You Score High on the ASVAB for Better Military Placement 7th Edition If you're seeking a high ASVAB score for a better position within the military or looking to get the minimum required score for military acceptance REA has Your Total ASVAB Solution Prepared by an educational testing expert each comprehensive review chapter covers all the skills tested on the ASVAB including communication arithmetic and technical skills Each chapter includes practice drills with answer explanations ASVAB test tips and subject specific sidebars that boost your knowledge The book contains a diagnostic test plus three full length practice tests that replicate the actual format and structure of the ASVAB so you can practice for real before test day Detailed explanations of answers are included for every test question allowing you to pinpoint your strengths and weaknesses and focus on areas in need of further study Our complete ASVAB test prep package also offers an 8 week study schedule and test taking strategies for succeeding on the exam This civil service vocational test prep is perfect for individuals seeking a high ASVAB score for a better position within the military and for those seeking to get the minimum required score for military acceptance It explains how military and other career counselors use the ASVAB Your Total ASVAB Solution is also helpful for individuals taking the ASVAB as a diagnostic test for further education or career planning More than 40 million people have taken the ASVAB since 1968 If you're next in line be prepared with Your Total ASVAB Solution

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The Case Against Fluoride Paul Connett,James Beck,Spedding Micklem,2010-09-29 In the case of water fluoridation the chemicals used to fluoridate the water that more than 180 million people drink each day are not pharmaceutical grade but rather hazardous waste products of the phosphate fertilizer industry it is illegal to dump them into rivers and lakes or release them into the atmosphere And water fluoridation is a prime example of one of the worst medical practices possible forced medication with no control over the dose or who gets it Perhaps most shocking of all it is not subject to any federal regulation At once painstakingly documented and also highly readable The Case Against Fluoride brings new research to light including links between fluoride and harm to the brain bones and kidneys and argues that while there is possible value in topical applications like brushing your teeth with fluoride toothpaste the evidence that swallowing fluoride reduces tooth decay is surprisingly weak

Comprehensive Textbook of Community Health Nursing Including Environmental Science (Two Volume Set), First Edition - E-Book Neerja Sood,Sakshi Chaturvedi,2025-04-07 This book has been written with student nurses learning needs in mind and it fully covers the Indian Nursing Council s new revised syllabus for nursing degree and diploma courses Although it primarily intends to cater to the curriculum demands of BSc Nursing and GNM students it will also be extremely valuable for students of various other streams studying community health courses MLHP CCH and MPH courses The main principles have been conveyed in clear terms from the students perspective The global and national health scenario and community health nursing principles are considered while explaining the role of community health nurses Designed to meet the curricular needs of student nurses Relevant case studies and examples included Figures tables and pictures are placed in order to enhance logical thinking Recent research work in the relevant field is included in the content Contains short answer and long answer exam oriented questions at the end of chapters Provides additional multiple choice questions to help students have a firm grasp on the subject

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