

Use the chart to determine the heat of formation for each reaction.
Use formula $\Delta H_{rxn} = \sum \Delta H_{f, products} - \sum \Delta H_{f, reactants}$

[illegible]

- $$2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}_{(l)}$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [2\text{H}_2\text{O}] - [2\text{H}_2 + \text{O}_2]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [2(-285.8)] - [2(0) + 0]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = -571.6 \text{ kJ}$$
- $$\text{CH}_4 + 2\text{O}_2 \rightarrow 2\text{H}_2\text{O}_{(g)} + \text{CO}_2$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [2\text{H}_2\text{O} + \text{CO}_2] - [\text{CH}_4 + 2\text{O}_2]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [2(-285.8) + (-393.5)] - [-74.8 + 2(0)]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = -890.3 \text{ kJ}$$
- $$\text{CuCl}_2 + 2\text{Ag}_{(s)} \rightarrow \text{Cu}_{(s)} + 2\text{AgCl} \quad (\Delta H^{\circ}, \text{CuCl}_2 = -215.6 \text{ kJ/mol})$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [\text{Cu}_{(s)} + 2\text{AgCl}] - [\text{CuCl}_2 + 2\text{Ag}_{(s)}]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [0 + 2(-127)] - [-215.6 + 2(0)]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = -38.4 \text{ kJ}$$
- $$\text{NaOH} + \text{HCl} \rightarrow \text{H}_2\text{O}_{(l)} + \text{NaCl}$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [\text{H}_2\text{O}_{(l)} + \text{NaCl}] - [\text{NaOH} + \text{HCl}]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [-285.8 + (-411.0)] - [-426.6 + (-92.3)]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = -177.9 \text{ kJ}$$
- $$2\text{NO} + \text{O}_2 \rightarrow 2\text{NO}_2$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [2\text{NO}_2] - [2\text{NO} + \text{O}_2]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [2(33.9)] - [2(90.4) + 0]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = -113.0 \text{ kJ}$$
- $$\text{N}_2\text{O} + \text{NO}_2 \rightarrow 3\text{NO} \quad (\Delta H^{\circ}, \text{N}_2\text{O} = 82.6 \text{ kJ/mol})$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [3\text{NO}] - [\text{N}_2\text{O} + \text{NO}_2]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [3(90.4)] - [82.6 + 33.9]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = 154.7 \text{ kJ}$$
- $$\text{CaC}_2 + 2\text{H}_2\text{O}_{(l)} \rightarrow \text{Ca}(\text{OH})_2 + \text{C}_2\text{H}_2 \quad (\Delta H^{\circ}, \text{CaC}_2 = -62.8 \text{ kJ/mol})$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [\text{Ca}(\text{OH})_2 + \text{C}_2\text{H}_2] - [\text{CaC}_2 + 2\text{H}_2\text{O}_{(l)}]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = [-986.6 + 226.7] - [-62.8 + 2(-285.8)]$$

$$\Delta^{\circ}\text{H}_{\text{rxn}} = -125.5 \text{ kJ}$$

Heat Of Formation Worksheet With Answers

**Arthur E. Morris, Gordon Geiger, H. Alan
Fine**



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modeling techniques on process safety applications It is a valuable resource for readers interested in theoretical simulations and or computer simulations of hazardous scenarios As multi scale modeling is a computational technique for solving problems involving multiple scales such as how a flammable vapor cloud might behave if ignited this book provides information on the fundamental topics of toxic fire and air explosion modeling as well as modeling jet and pool fires using computational fluid dynamics The book goes on to cover nanomaterial toxicity QPSR analysis on relation of chemical structure to flash point molecular structure and burning velocity first principle studies of reactive chemicals water and air reactive chemicals and dust explosions Chemical and process safety professionals as well as faculty and graduate researchers will benefit from the detailed coverage provided in this book Provides the only comprehensive source addressing the use of multiscale modeling in the context of process safety Bridges multiscale modeling with process safety enabling the reader to understand mapping between problem detail and effective usage of resources Presents an overall picture of addressing safety problems in all levels of modeling and the latest approaches to each in the field Features worked out examples case studies and a question bank to aid understanding and involvement for the reader

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Pm Science Practice P5/6 , **The Science Hub-TM** Preetika Sawhney, Archana Sashi Kumar, Neha Jindal, Gautam Bindal, Shalini Samadhiya and Tripti Mehta, A Book on Science Teacher Manual The ebook version does not contain CD

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