

A 20 kg object moving at 24 m/s slows to 3 m/s due to friction.

A) How much energy is converted to heat?

The loss of  $E_k$  turns to  $Q$ .

$$E_{ki} = \frac{1}{2}(20)(24)^2 = 5760 \text{ initial}$$

$$E_{kf} = \frac{1}{2}(20)(3)^2 = 90 \text{ Final}$$

$$\Delta E_k = 5670 \text{ J}$$

B) If the box absorbs three-fourths of the energy, how much does it gain?

$$5670(.75) = 4253 \text{ J}$$

C) Is the box or the floor a better conductor? ~~box~~ - because it absorbs more of the  $Q$ .

D) If the box is made up copper and originally at  $22^\circ\text{C}$ , what is its final temperature after it stops?

$$Q = mc\Delta T = 20(387)(T_f - 22)$$

$$4253 = 7740(T_f - 22)$$

$$.549 = T_f - 22$$

$$T_f = 22.549^\circ\text{C}$$

If a piece of wood and a piece of metal of equal mass are laying on a table in the sun.

A) After a while, which one will be hotter?

metal

B) Why?

conductor - lower  $C_p$

A 250 g iron bolt at  $70^\circ\text{C}$  is dropped into an 100g aluminum calorimeter of containing 200 g of water.

If the water and calorimeter are at room temperature ( $22^\circ\text{C}$ ) to begin with, find the final temperature when they come to thermal equilibrium? ~~see below~~

$$m_{\text{iron}} = .25 \text{ kg}$$

$$m_{\text{water}} = .2 \text{ kg}$$

$$m_{\text{cal}} = .1 \text{ kg}$$

$$-Q_{\text{hot}} = Q_{\text{cold water}} + Q_{\text{cold calorimeter}}$$

$$-m c \Delta T = m c \Delta T + m c \Delta T$$

$$-.25(448)(T_f - 70) = .2(4184)(T_f - 22) + .1(899)(T_f - 22)$$

$$= (T_f - 22)(837.2 + 89.9)$$

$$-112(T_f - 70) = (T_f - 22)927.1$$

$$-112T_f + 7840 = 927.1T_f - 20396$$

$$2.8 \times 10^4 = 1039T_f$$

$$26.9^\circ\text{C} = T_f$$

\* - we put a negative on the left side because it is already negative (it is losing  $Q$ ).

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**Jianjun Gao**



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Heat And Thermo 1 Answer Key Stephen Murray**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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### **Table of Contents Heat And Thermo 1 Answer Key Stephen Murray**

1. Understanding the eBook Heat And Thermo 1 Answer Key Stephen Murray
  - The Rise of Digital Reading Heat And Thermo 1 Answer Key Stephen Murray
  - Advantages of eBooks Over Traditional Books
2. Identifying Heat And Thermo 1 Answer Key Stephen Murray
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Heat And Thermo 1 Answer Key Stephen Murray
  - User-Friendly Interface
4. Exploring eBook Recommendations from Heat And Thermo 1 Answer Key Stephen Murray
  - Personalized Recommendations
  - Heat And Thermo 1 Answer Key Stephen Murray User Reviews and Ratings
  - Heat And Thermo 1 Answer Key Stephen Murray and Bestseller Lists

5. Accessing Heat And Thermo 1 Answer Key Stephen Murray Free and Paid eBooks
  - Heat And Thermo 1 Answer Key Stephen Murray Public Domain eBooks
  - Heat And Thermo 1 Answer Key Stephen Murray eBook Subscription Services
  - Heat And Thermo 1 Answer Key Stephen Murray Budget-Friendly Options
6. Navigating Heat And Thermo 1 Answer Key Stephen Murray eBook Formats
  - ePub, PDF, MOBI, and More
  - Heat And Thermo 1 Answer Key Stephen Murray Compatibility with Devices
  - Heat And Thermo 1 Answer Key Stephen Murray Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Heat And Thermo 1 Answer Key Stephen Murray
  - Highlighting and Note-Taking Heat And Thermo 1 Answer Key Stephen Murray
  - Interactive Elements Heat And Thermo 1 Answer Key Stephen Murray
8. Staying Engaged with Heat And Thermo 1 Answer Key Stephen Murray
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Heat And Thermo 1 Answer Key Stephen Murray
9. Balancing eBooks and Physical Books Heat And Thermo 1 Answer Key Stephen Murray
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Heat And Thermo 1 Answer Key Stephen Murray
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Heat And Thermo 1 Answer Key Stephen Murray
  - Setting Reading Goals Heat And Thermo 1 Answer Key Stephen Murray
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Heat And Thermo 1 Answer Key Stephen Murray
  - Fact-Checking eBook Content of Heat And Thermo 1 Answer Key Stephen Murray
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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