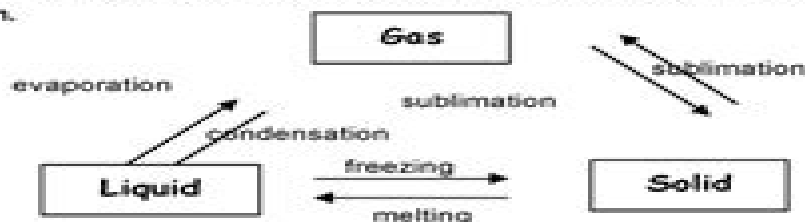


Chemistry & Matter

What is chemistry? *The study of matter and the changes it undergoes.*

1. Matter is anything that has mass and occupies space. Matter can exist in any one of three states: *solid (s), liquid (l) and gas (g).*

Add the following labels in the appropriate places: evaporation, condensation, melting, freezing, sublimation.

Particle Theory of Matter

1. All matter is made up of tiny particles.
2. All particles of a pure substance are identical.
3. The space between particles is large compared to the sizes of the particles themselves.
4. Particles of matter are always moving.
5. There are forces of attraction that exist between particles.

Describing and Classifying Matter

1. Distinguish between a physical change and chemical change.

Physical change- change that affects the physical appearance but not its composition (change of state, form, dissolving)

Chemical change-occurs when a new substance with new physical and chemical properties are formed. Clues that a chemical change occurred are: new colour, gas formed, precipitate, or heat/light is given off or heat is absorbed.

2. Classify each situation as either a physical change or a chemical change. Explain your reasoning.
- (a) Rose bush from a seed that you have planted and nourished - Chemical change
 - (b) A green coating forms on a copper statue when the statue is exposed to air - Chemical change
 - (c) Your sweat evaporates to help balance your body temperature - Physical change
 - (d) Frost forms on the inside of a freezer - Physical change
 - (e) Salt is added to clear chicken broth - Physical change
 - (f) Your body breaks down food you eat to provide energy for your body's cells. - Chemical change
 - (g) Juice crystals dissolve in water - Physical change
 - (h) An ice-cream cone melts on a hot day - Physical change

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1. Understanding the eBook Grade 10 Chemistry Review With Answers
 - The Rise of Digital Reading Grade 10 Chemistry Review With Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 10 Chemistry Review With Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 10 Chemistry Review With Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 10 Chemistry Review With Answers
 - Personalized Recommendations
 - Grade 10 Chemistry Review With Answers User Reviews and Ratings
 - Grade 10 Chemistry Review With Answers and Bestseller Lists
5. Accessing Grade 10 Chemistry Review With Answers Free and Paid eBooks
 - Grade 10 Chemistry Review With Answers Public Domain eBooks
 - Grade 10 Chemistry Review With Answers eBook Subscription Services

- Grade 10 Chemistry Review With Answers Budget-Friendly Options
- 6. Navigating Grade 10 Chemistry Review With Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 10 Chemistry Review With Answers Compatibility with Devices
 - Grade 10 Chemistry Review With Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 10 Chemistry Review With Answers
 - Highlighting and Note-Taking Grade 10 Chemistry Review With Answers
 - Interactive Elements Grade 10 Chemistry Review With Answers
- 8. Staying Engaged with Grade 10 Chemistry Review With Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 10 Chemistry Review With Answers
- 9. Balancing eBooks and Physical Books Grade 10 Chemistry Review With Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 10 Chemistry Review With Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 10 Chemistry Review With Answers
 - Setting Reading Goals Grade 10 Chemistry Review With Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 10 Chemistry Review With Answers
 - Fact-Checking eBook Content of Grade 10 Chemistry Review With Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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