

HEAT & TEMPERATURE



READINGS
ACTIVITIES
ASSESSMENTS

ALBERTA — GRADE 7 — SCIENCE

Three sample worksheets are displayed, overlapping each other. The worksheets are titled "Heat vs Temperature", "Boiling Curve - Boiling's Temperature", and "Heat vs Temperature".

The "Boiling Curve - Boiling's Temperature" worksheet includes a table with the following data:

Time	Temperature
0	100 °C
10	100 °C
20	100 °C
30	100 °C
40	100 °C
50	100 °C
60	100 °C
70	100 °C
80	100 °C
90	100 °C
100	100 °C

The "Heat vs Temperature" worksheet includes a diagram of a pot of water on a stove with flames, and a thermometer.

Heat vs Temperature

Heat is a form of energy that is transferred from one object to another. It is measured in Joules (J). Temperature is a measure of the average kinetic energy of the particles in a substance. It is measured in degrees Celsius (°C).

When we heat a substance, we are adding energy to it. This energy is transferred to the particles in the substance, which then move faster. This causes the temperature of the substance to rise.

When we cool a substance, we are removing energy from it. This energy is transferred from the particles in the substance, which then move slower. This causes the temperature of the substance to fall.

Heat and temperature are related, but they are not the same thing. Heat is a form of energy, while temperature is a measure of the average kinetic energy of the particles in a substance.

For example, if you have a pot of water on a stove, the heat from the stove is transferred to the water. This causes the temperature of the water to rise. If you have a thermometer in the water, it will show the temperature of the water.

Heat and temperature are both measured in Joules (J). However, heat is measured in Joules (J), while temperature is measured in degrees Celsius (°C).

Heat and temperature are both important concepts in science. They are used to describe the energy of a system and the temperature of a substance.

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Heat And Temperature Grade 7

**Gerhard Welsch,Rodney Boyer,E. W.
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