

# Chemistry Spring Final Study Guide

### Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
<p><b>Unit 10: Acids &amp; Bases</b></p> <p>Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below.</p> <p>Label the y-axis.</p> <p>Label the x-axis.</p> <p>1. A substance of the particles.</p> <p>2. A substance of the particles.</p> <p>3. A substance of the particles.</p> <p>4. The gas is in.</p> <p>5. The volume.</p>	<p>1. A substance of the particles.</p> <p>2. A substance of the particles.</p> <p>3. A substance of the particles.</p> <p>4. The gas is in.</p> <p>5. The volume.</p>

**Unit 10: Acids & Bases**

Write the definition:

- Acid
- Base
- Indicator
- Titration
- Endpoint
- Equivalence
- Titration

Use the equation that:

Write the correct formula:

1. Write the correct formula of the acid.

2. Write the correct formula of the base.

3. Write the correct formula of the salt.

### Chemistry Spring Final Study Guide

Name: \_\_\_\_\_

**Unit 1: Measurement & Analysis**

Mark the best answer from those provided to each question.

Mark the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers:

1. 12.3456789  $\rightarrow$  \_\_\_\_\_ sig figs

2. 12.3456789  $\rightarrow$  \_\_\_\_\_ sig figs

3. 12.3456789  $\rightarrow$  \_\_\_\_\_ sig figs

4. 12.3456789  $\rightarrow$  \_\_\_\_\_ sig figs

Convert the following into Scientific Notation:

1. 12.3456789  $\rightarrow$  \_\_\_\_\_

2. 12.3456789  $\rightarrow$  \_\_\_\_\_

3. 12.3456789  $\rightarrow$  \_\_\_\_\_

4. 12.3456789  $\rightarrow$  \_\_\_\_\_

Convert the following into Decimal Notation:

1. 12.3456789  $\rightarrow$  \_\_\_\_\_

2. 12.3456789  $\rightarrow$  \_\_\_\_\_

3. 12.3456789  $\rightarrow$  \_\_\_\_\_

4. 12.3456789  $\rightarrow$  \_\_\_\_\_

**Unit 2: Matter & Energy**

Mark the best answer from those provided to each question.

Mark the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers:

1. 12.3456789  $\rightarrow$  \_\_\_\_\_ sig figs

2. 12.3456789  $\rightarrow$  \_\_\_\_\_ sig figs

3. 12.3456789  $\rightarrow$  \_\_\_\_\_ sig figs

4. 12.3456789  $\rightarrow$  \_\_\_\_\_ sig figs

Convert the following into Scientific Notation:

1. 12.3456789  $\rightarrow$  \_\_\_\_\_

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3. 12.3456789  $\rightarrow$  \_\_\_\_\_

4. 12.3456789  $\rightarrow$  \_\_\_\_\_

Convert the following into Decimal Notation:

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2. 12.3456789  $\rightarrow$  \_\_\_\_\_

3. 12.3456789  $\rightarrow$  \_\_\_\_\_

4. 12.3456789  $\rightarrow$  \_\_\_\_\_

**Unit 3: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $CO_2$

3.  $CH_4$

4.  $NaCl$

5.  $KNO_3$

6.  $Ca(OH)_2$

7.  $Al_2(SO_4)_3$

8.  $Fe_2O_3$

9.  $Na_2CO_3$

10.  $H_2SO_4$

11.  $NaOH$

12.  $CaCl_2$

13.  $Na_2SO_4$

14.  $Na_2CO_3$

15.  $Na_2SO_4$

16.  $Na_2CO_3$

17.  $Na_2SO_4$

18.  $Na_2CO_3$

19.  $Na_2SO_4$

20.  $Na_2CO_3$

21.  $Na_2SO_4$

22.  $Na_2CO_3$

23.  $Na_2SO_4$

24.  $Na_2CO_3$

25.  $Na_2SO_4$

26.  $Na_2CO_3$

27.  $Na_2SO_4$

28.  $Na_2CO_3$

29.  $Na_2SO_4$

30.  $Na_2CO_3$

31.  $Na_2SO_4$

32.  $Na_2CO_3$

33.  $Na_2SO_4$

34.  $Na_2CO_3$

35.  $Na_2SO_4$

36.  $Na_2CO_3$

37.  $Na_2SO_4$

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39.  $Na_2SO_4$

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95.  $Na_2SO_4$

96.  $Na_2CO_3$

97.  $Na_2SO_4$

98.  $Na_2CO_3$

99.  $Na_2SO_4$

100.  $Na_2CO_3$

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**A Loxley**



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