

## Stoichiometry Worksheet 2: Percent Yield

For each of the problems below:

- Write the balanced chemical equation
- Identify the given (with units) and what you want to find (with units)
- Show set up with units. Check sig figs, give final answer with units and label.

1. Using the Hoffman apparatus for electrolysis, a chemist decomposes 36 g of water into its gaseous elements. How many grams of hydrogen gas should she get (theoretical yield)?



Before 2 mol 0 mol 0 mol

Change -2 mol +2 mol 0 mol

After 0 mol 2 mol 0 mol

$$36 \text{ g H}_2\text{O} \times \frac{1 \text{ mol H}_2\text{O}}{18.0 \text{ g H}_2\text{O}} = 2.0 \text{ mol H}_2\text{O} \times \frac{2 \text{ mol H}_2}{2 \text{ mol H}_2\text{O}} = 2 \text{ mol H}_2 \times \frac{2.0 \text{ g H}_2}{1 \text{ mol H}_2} = 4.0 \text{ g H}_2$$

2. Recall that liquid sodium reacts with chlorine gas to produce sodium chloride. You want to produce 581 g of sodium chloride. How many grams of sodium are needed?



Before 9.94 mol xs mol 0 mol

Change -9.94 mol 0 mol +9.94 mol

After 0 mol 0 mol 9.94 mol

$$581 \text{ g NaCl} \times \frac{1 \text{ mol NaCl}}{58.5 \text{ g NaCl}} = 9.93 \text{ mol NaCl} \times \frac{2 \text{ mol Na}}{2 \text{ mol NaCl}} = 9.93 \text{ mol Na} \times \frac{23.0 \text{ g Na}}{1 \text{ mol Na}} = 228 \text{ g Na}$$

3. You eat 180.0 g of glucose (90 M&Ms). If glucose,  $\text{C}_6\text{H}_{12}\text{O}_6$ , reacts with oxygen gas to produce carbon dioxide and water, how many grams of oxygen will you have to breathe in to burn the glucose?



Before 1.0 mol xs mol 0 mol 0 mol

Change -1.0 mol -6.0 mol +6.0 mol +6.0 mol

After 0 mol xs mol 0 mol 0 mol

$$180.0 \text{ g C}_6\text{H}_{12}\text{O}_6 \times \frac{1 \text{ mol C}_6\text{H}_{12}\text{O}_6}{180 \text{ g C}_6\text{H}_{12}\text{O}_6} = 1.00 \text{ mol C}_6\text{H}_{12}\text{O}_6 \times \frac{6 \text{ mol O}_2}{1 \text{ mol C}_6\text{H}_{12}\text{O}_6} = 6.00 \text{ mol O}_2$$

$$6.0 \text{ mol O}_2 \times \frac{32.0 \text{ g O}_2}{1 \text{ mol O}_2} = 192.0 \text{ g O}_2$$

# Guide Chemistry Stoichiometry Answer Key

**Martha Mackin**



## **Guide Chemistry Stoichiometry Answer Key:**

## **Guide Chemistry Stoichiometry Answer Key** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Guide Chemistry Stoichiometry Answer Key**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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### **Table of Contents Guide Chemistry Stoichiometry Answer Key**

1. Understanding the eBook Guide Chemistry Stoichiometry Answer Key
  - The Rise of Digital Reading Guide Chemistry Stoichiometry Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Guide Chemistry Stoichiometry Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Guide Chemistry Stoichiometry Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Guide Chemistry Stoichiometry Answer Key
  - Personalized Recommendations
  - Guide Chemistry Stoichiometry Answer Key User Reviews and Ratings
  - Guide Chemistry Stoichiometry Answer Key and Bestseller Lists

5. Accessing Guide Chemistry Stoichiometry Answer Key Free and Paid eBooks
  - Guide Chemistry Stoichiometry Answer Key Public Domain eBooks
  - Guide Chemistry Stoichiometry Answer Key eBook Subscription Services
  - Guide Chemistry Stoichiometry Answer Key Budget-Friendly Options
6. Navigating Guide Chemistry Stoichiometry Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Guide Chemistry Stoichiometry Answer Key Compatibility with Devices
  - Guide Chemistry Stoichiometry Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Guide Chemistry Stoichiometry Answer Key
  - Highlighting and Note-Taking Guide Chemistry Stoichiometry Answer Key
  - Interactive Elements Guide Chemistry Stoichiometry Answer Key
8. Staying Engaged with Guide Chemistry Stoichiometry Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Guide Chemistry Stoichiometry Answer Key
9. Balancing eBooks and Physical Books Guide Chemistry Stoichiometry Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Guide Chemistry Stoichiometry Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Guide Chemistry Stoichiometry Answer Key
  - Setting Reading Goals Guide Chemistry Stoichiometry Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Guide Chemistry Stoichiometry Answer Key
  - Fact-Checking eBook Content of Guide Chemistry Stoichiometry Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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