

# Periodic Table of the Elements

# Periodic Table of the Elements

| Atomic Number | Symbol | Name | Common State | Common Configuration |
|---------------|--------|------|--------------|----------------------|
|               |        |      |              |                      |

Colors represent groups of elements with similar properties

Solid, Liquid or Gas

|     |     |     |     |     |     |     |     |     |     |
|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| 3   | 4   | 5   | 6   | 7   | 8   | 9   | 10  | 11  | 12  |
| 108 | 108 | 108 | 108 | 108 | 108 | 108 | 108 | 108 | 108 |
| 108 | 108 | 108 | 108 | 108 | 108 | 108 | 108 | 108 | 108 |

|          |           |                   |               |            |            |         |           |            |              |
|----------|-----------|-------------------|---------------|------------|------------|---------|-----------|------------|--------------|
| 21       | 22        | 23                | 24            | 25         | 26         | 27      | 28        | 29         | 30           |
| Scandium | Titanium  | Vanadium          | Chromium      | Manganese  | Iron       | Cobalt  | Nickel    | Copper     | Zinc         |
| Scandium | Titanium  | Vanadium          | Chromium      | Manganese  | Iron       | Cobalt  | Nickel    | Copper     | Zinc         |
| 39       | 40        | 41                | 42            | 43         | 44         | 45      | 46        | 47         | 48           |
| Yttrium  | Zirconium | Niobium           | Molybdenum    | Technetium | Ruthenium  | Rhodium | Palladium | Silver     | Cadmium      |
| Yttrium  | Zirconium | Niobium           | Molybdenum    | Technetium | Ruthenium  | Rhodium | Palladium | Silver     | Cadmium      |
| 55       | 56        | 57-71             | 72            | 73         | 74         | 75      | 76        | 77         | 78           |
| Cesium   | Barium    | Lanthanide Series | Hafnium       | Tantalum   | Tungsten   | Rhenium | Osmium    | Iridium    | Platinum     |
| Cesium   | Barium    | Lanthanide Series | Hafnium       | Tantalum   | Tungsten   | Rhenium | Osmium    | Iridium    | Platinum     |
| 87       | 88        | 89-103            | 104           | 105        | 106        | 107     | 108       | 109        | 110          |
| Francium | Radium    | Actinide Series   | Rutherfordium | Dubnium    | Seaborgium | Bohrium | Hassium   | Meitnerium | Darmstadtium |
| Francium | Radium    | Actinide Series   | Rutherfordium | Dubnium    | Seaborgium | Bohrium | Hassium   | Meitnerium | Darmstadtium |

|          |           |          |           |          |         |
|----------|-----------|----------|-----------|----------|---------|
| 13       | 14        | 15       | 16        | 17       | 18      |
| Boron    | Carbon    | Nitrogen | Oxygen    | Fluorine | Neon    |
| Boron    | Carbon    | Nitrogen | Oxygen    | Fluorine | Neon    |
| 31       | 32        | 33       | 34        | 35       | 36      |
| Gallium  | Germanium | Arsenic  | Selenium  | Bromine  | Krypton |
| Gallium  | Germanium | Arsenic  | Selenium  | Bromine  | Krypton |
| 49       | 50        | 51       | 52        | 53       | 54      |
| Indium   | Tin       | Antimony | Tellurium | Iodine   | Xenon   |
| Indium   | Tin       | Antimony | Tellurium | Iodine   | Xenon   |
| 81       | 82        | 83       | 84        | 85       | 86      |
| Thallium | Lead      | Bismuth  | Polonium  | Astatine | Radon   |
| Thallium | Lead      | Bismuth  | Polonium  | Astatine | Radon   |
| 113      | 114       | 115      | 116       | 117      | 118     |
| Uut      | Fl        | Uup      | Lv        | Uus      | Uuo     |
| Uut      | Fl        | Uup      | Lv        | Uus      | Uuo     |

|                   |                                                                         |                                                                       |                                                                                            |                                                                                      |                                                                                        |                                                                                        |                                                                        |                                                                                          |                                                                        |                                                                           |                                                                           |                                                                       |                                                                            |                                                                          |                                                                                           |
|-------------------|-------------------------------------------------------------------------|-----------------------------------------------------------------------|--------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------|------------------------------------------------------------------------|------------------------------------------------------------------------------------------|------------------------------------------------------------------------|---------------------------------------------------------------------------|---------------------------------------------------------------------------|-----------------------------------------------------------------------|----------------------------------------------------------------------------|--------------------------------------------------------------------------|-------------------------------------------------------------------------------------------|
| Lanthanide Series | 57<br>La<br>Lanthanum<br>138.91<br>[Xe] 5d <sup>1</sup> 6s <sup>2</sup> | 58<br>Ce<br>Cerium<br>140.12<br>[Xe] 4f <sup>1</sup> 6s <sup>2</sup>  | 59<br>Pr<br>Praseodymium<br>140.91<br>[Xe] 4f <sup>3</sup> 6s <sup>2</sup>                 | 60<br>Nd<br>Neodymium<br>144.24<br>[Xe] 4f <sup>4</sup> 6s <sup>2</sup>              | 61<br>Pm<br>Promethium<br>[145]<br>[Xe] 4f <sup>5</sup> 6s <sup>2</sup>                | 62<br>Sm<br>Samarium<br>150.36<br>[Xe] 4f <sup>6</sup> 6s <sup>2</sup>                 | 63<br>Eu<br>Europium<br>151.96<br>[Xe] 4f <sup>7</sup> 6s <sup>2</sup> | 64<br>Gd<br>Gadolinium<br>157.25<br>[Xe] 4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup> | 65<br>Tb<br>Terbium<br>158.93<br>[Xe] 4f <sup>9</sup> 6s <sup>2</sup>  | 66<br>Dy<br>Dysprosium<br>162.50<br>[Xe] 4f <sup>10</sup> 6s <sup>2</sup> | 67<br>Ho<br>Holmium<br>164.93<br>[Xe] 4f <sup>11</sup> 6s <sup>2</sup>    | 68<br>Er<br>Erbium<br>167.26<br>[Xe] 4f <sup>12</sup> 6s <sup>2</sup> | 69<br>Tm<br>Thulium<br>168.93<br>[Xe] 4f <sup>13</sup> 6s <sup>2</sup>     | 70<br>Yb<br>Ytterbium<br>173.05<br>[Xe] 4f <sup>14</sup> 6s <sup>2</sup> | 71<br>Lu<br>Lutetium<br>174.97<br>[Xe] 4f <sup>14</sup> 5d <sup>1</sup> 6s <sup>2</sup>   |
| Actinide Series   | 88<br>Ac<br>Actinium<br>[227]<br>[Rn] 6d <sup>1</sup> 7s <sup>2</sup>   | 89<br>Th<br>Thorium<br>232.04<br>[Rn] 6d <sup>2</sup> 7s <sup>2</sup> | 90<br>Pa<br>Protactinium<br>231.04<br>[Rn] 5f <sup>2</sup> 6d <sup>1</sup> 7s <sup>2</sup> | 91<br>U<br>Uranium<br>238.03<br>[Rn] 5f <sup>3</sup> 6d <sup>1</sup> 7s <sup>2</sup> | 92<br>Np<br>Neptunium<br>[237]<br>[Rn] 5f <sup>4</sup> 6d <sup>1</sup> 7s <sup>2</sup> | 93<br>Pu<br>Plutonium<br>[244]<br>[Rn] 5f <sup>6</sup> 6d <sup>1</sup> 7s <sup>2</sup> | 94<br>Am<br>Americium<br>[243]<br>[Rn] 5f <sup>7</sup> 7s <sup>2</sup> | 95<br>Cm<br>Curium<br>[247]<br>[Rn] 5f <sup>7</sup> 6d <sup>1</sup> 7s <sup>2</sup>      | 96<br>Bk<br>Berkelium<br>[247]<br>[Rn] 5f <sup>9</sup> 7s <sup>2</sup> | 97<br>Cf<br>Californium<br>[251]<br>[Rn] 5f <sup>10</sup> 7s <sup>2</sup> | 98<br>Es<br>Einsteinium<br>[252]<br>[Rn] 5f <sup>11</sup> 7s <sup>2</sup> | 99<br>Fm<br>Fermium<br>[257]<br>[Rn] 5f <sup>12</sup> 7s <sup>2</sup> | 100<br>Md<br>Mendelevium<br>[258]<br>[Rn] 5f <sup>13</sup> 7s <sup>2</sup> | 101<br>No<br>Nobelium<br>[259]<br>[Rn] 5f <sup>14</sup> 7s <sup>2</sup>  | 102<br>Lr<br>Lawrencium<br>[260]<br>[Rn] 5f <sup>14</sup> 6d <sup>1</sup> 7s <sup>2</sup> |

|              |                |                  |            |           |          |          |           |            |          |
|--------------|----------------|------------------|------------|-----------|----------|----------|-----------|------------|----------|
| Alkali Metal | Alkaline Earth | Transition Metal | Post Metal | Metalloid | Nonmetal | Halogens | Noble Gas | Lanthanide | Actinide |
|--------------|----------------|------------------|------------|-----------|----------|----------|-----------|------------|----------|

# Guide To The Periodic Table

**Albert Stwertka**



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