

5 Find out about inspections

At least ten days before work starts, the PCA must inform you about the inspections required and you must inform your builder. The builder will make arrangements with the PCA for inspections to be carried out as needed. The PCA may not be able to issue certificates for completed work if inspections are not undertaken.

6 Obtain an occupation certificate

Before you can occupy the building, you must apply for and obtain an occupation certificate from your PCA.

The PCA will inspect the completed building work to determine if it is suitable to occupy. If satisfied, they will issue an occupation certificate.

You can appeal to the Land and Environment Court if the council refuses to issue a final occupation certificate (or construction certificate), or if no decision is made after a certain period. You cannot appeal against the decision of an accredited certifier to refuse an application for a certificate.

If you believe your PCA is unreasonably delaying the release of your certificate or approval you should, in the first instance, discuss this with the PCA. If you are not satisfied after that, contact the Board for advice and assistance.

Not all accredited private certifiers can issue all of the above certificates or conduct mandatory inspections. Their authority to issue these certificates depends upon their level of accreditation. Before you engage a certifier, you should check their level of accreditation at www.bpb.nsw.gov.au.

Councils can issue all of the certificates mentioned above.

Find out more

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Tel: (02) 9695 5950
Fax: (02) 9695 5949
Email: info@bpb.nsw.gov.au
www.bpb.nsw.gov.au

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working with a PCA in
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Guide to the building approvals process

A six-step guide to the approvals required for anyone intending to build



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Guide To The Building Building Professionals Board

JA Banks



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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web the pressure volume temperature pvt fluids lab in ku s chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

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web volume pressure kPa linear pressure kPa volume mL pressure kPa in part b we observed the relationship between temperature and pressure the pressure of the 125 mL flask was at 101 36 kPa when the temperature was at 23 2 °C

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

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