

Chapter 21: NUCLEAR CHEMISTRY - Study Guide

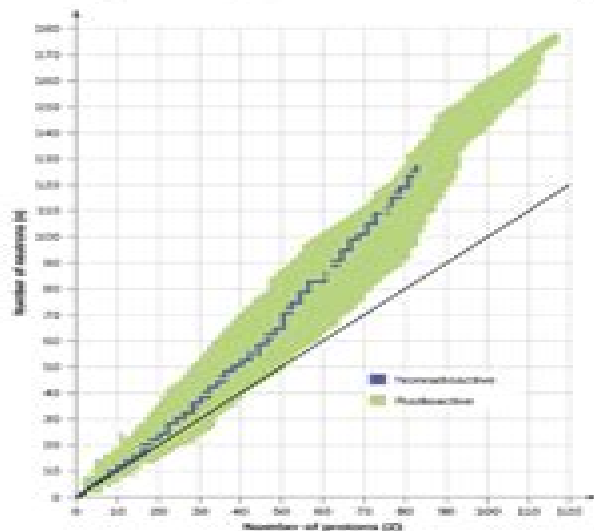
CHEM 1412 Chemistry OpenStax

Answer

1. A nuclide refers to _____.
2. Complete the following table:

	<i>symbol</i>	<i>definition</i>
Mass number		#protons + # neutrons
Atomic number	<i>Z</i>	

3. In the graph below (Figure 21.2 in OpenStax), the blue data points represent stable nuclides. The straight line represents a 1:1 neutron:proton ratio.



Heavy nuclides need a _____
(*higher/lower*) ratio of neutrons to protons to be stable.

Are even or odd numbers of protons and neutrons more likely to be stable? _____

What are magic numbers?

4. Which type of radioactive decay changes the mass number AND the atomic number?

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