

CHAPTER 6 REVIEW*Chemical Bonding***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ lattice energy _____	a. ionic bond
_____ bond energy _____	b. covalent bond
_____ enthalpy of vaporization _____	c. metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ ionic _____	a. H—F	_____ nonpolar covalent _____	d. H—H
_____ ionic _____	b. Na—Cl	_____ polar covalent _____	e. H—C
_____ polar covalent _____	c. H—O	_____ polar covalent _____	f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen atom and the unshared electron pair of a strongly electronegative atom such as O, N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction. Hydrogen bonds are generally intermolecular, while ionic and covalent bonds occur between ions or atoms respectively.

4. H₂S and H₂O have similar structures and their central atoms belong to the same group. Yet H₂S is a gas at room temperature and H₂O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond. Increased polarity in H₂O bonds means a stronger intermolecular attraction, making water a liquid at room temperature. Hydrogen bonding exists between water molecules, but not between hydrogen sulfide molecules.

Guided And Study Workbook Answers Covalent Bonding

Daniela Niemeyer



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Study Guide for The Human Body in Health & Disease - E-Book Kevin T. Patton, Frank B. Bell, Terry Thompson, Peggie L. Williamson, Eric L. Sun, 2023-03-02 Reinforce your understanding of the concepts in Patton's The Human Body in Health Disease Corresponding to the chapters in the text this study guide reviews essential medical terminology concepts and processes related to anatomy and physiology and explains how body systems function in health and disease Each chapter begins with a quick synopsis of the key points in the textbook chapter A variety of exercises make it easier to review and apply key concepts and labeling of anatomy drawings helps in learning anatomical terms and structures Know your Medical Terms feature helps familiarize you with the various word parts used in medical terminology and pairs nicely with the Language of Medicine word lists in the main text to reinforce medical terminology concepts that are key to understanding A P Brief synopsis of core textbook concepts provides a comprehensive review of essential content Crossword Puzzles and Word Finds help you master new vocabulary terms Application Questions ask you to make judgments based on the information in the chapter Diagrams and labeling exercises help reinforce where the structures of the body are located Matching and fill in the blank exercises aid in understanding chapter content Answers to exercises are provided in the back of the guide along with references to the appropriate textbook page and the textbook objective so that you can easily find where the material is presented Study tips in the Preface provide an overview of the most effective methods for learning and retaining information

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biology Sample Question Paper 2 This question is about the impact of potentially harmful chemicals and microorganisms. (a) (i). Salts that a plant needs, such as nitrates and phosphates, are ... Summary Notes - Topic 6.3 OCR (A) Biology A-Level The process occurs as following: • Nitrogen is first fixed by bacteria such as Rhizobium which live in the root nodules of leguminous plants such as pea plants. A level biology- enzymes A level biology- enzymes ... Explain how the following food preservation works: 1) Placing peas in boiling water for 1 minute then freezing them at -18 degrees. 2 ... ocr-a-level-biology-a-sb2-answers.pdf (e) Illuminated chloroplast produces oxygen; in light-dependent stage of photosynthesis; from photolysis of water; bacteria cluster where there is most oxygen; ... ocr a level biology nitrogen cycle Flashcards rhizobium as a nitrogen fixing bacteria. found in root nodules of leguminous plants such as peas and beans. nitrification definition. the process of converting ... The Nitrogen Cycle A2 OCR Biology Asking questions is a ... The Nitrogen Cycle A2 OCR Biology Asking questions is a sign of INTELLIGENCE ... bacteria) nitrogen fixing plant eg pea, clover bacteria. Nitrogen in the air ... 5.4.1 Plant Responses - 5.4.1 OCR bio notes Absciscic acid Inhibit seed germination and growth of stems. Ethene Promotes fruit ripening. The cell wall around a plant cell limits the cell's ability to divide ...