

## Fourth Grade Math Rubric

Math	1 No Progress  Unable to attempt to:	2 Limited Progress toward Grade Level Standard  Attempts but is unable to:	3 Progressing toward Grade  Inconsistently:	4 Meets Grade Level Standard  Consistently:	5 Exceeds Grade Level Standard  In addition to Level 4...
Uses the four operations with whole numbers to solve problems (4.OA.3)	<ul style="list-style-type: none"> <li>Solve word problems having whole number answers using the four operations.</li> </ul>	<ul style="list-style-type: none"> <li>Solve word problems having whole number answers using the four operations.</li> </ul>	<p>Solves multi-step word problems having whole number answers using:</p> <ul style="list-style-type: none"> <li>the four operations including the interpretation of remainders</li> <li>equations with a letter standing for the unknown quantity</li> <li>estimation to assess the reasonableness of an answer</li> </ul>	<p>Solves multi-step word problems having whole number answers using:</p> <ul style="list-style-type: none"> <li>the four operations including the interpretation of remainders</li> <li>equations with a letter standing for the unknown quantity</li> <li>estimation to assess the reasonableness of an answer</li> </ul>	<ul style="list-style-type: none"> <li>Fluently uses all the standard algorithms within each operation</li> </ul>

# Go Math 4th Grade Rubric

**William Tyler Smith**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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web allow the gaseous phase thermodynamics to be curve fitted to 300k about 142k lower than the normal boiling point 542k 2 of kerosene nevertheless in actual calculations involving liquid kerosene fuel the heat capacity and latent heat of vaporization of liquid kerosene 12 are used for

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web as a general trend vapor pressures of liquids at ambient temperatures increase with decreasing boiling points this is illustrated in the vapor pressure chart see right that shows graphs of the vapor pressures versus temperatures for a variety of liquids

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