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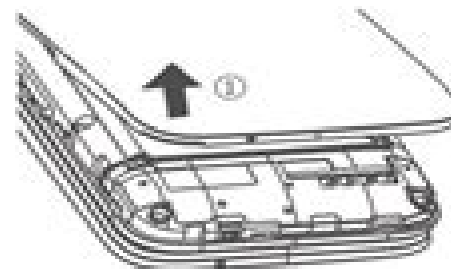
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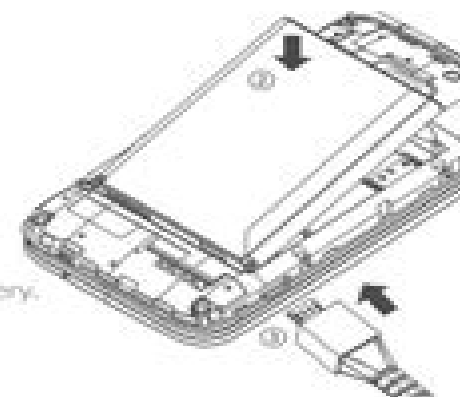
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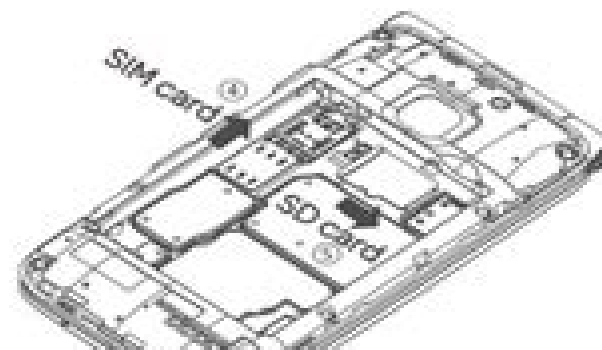


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User guide and FAQ are available online
www.alcatelone-touch.ca

Go Phone Manual

Patrick Vollmar



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web we looked at branched chain alkanes let s look at cyclo alkanes so this is a pretty funny dot structure here let s see how many carbons are in this triangle well of course there s one two and three carbons so if i were to draw what this molecule looks like if i were to draw all the atoms involved there d be three carbons like that

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three parallel lines between the two atoms and the bond angle between the atoms attached to them should

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web carbon 1 sp 180 carbon 2 sp 180 carbon 3 sp^2 120 carbon 4 sp^2 120 carbon 5 sp^3 109.5 chemically the alkynes are similar to the alkenes since the $C \equiv C$ functional group has two π bonds alkynes typically react even more readily and react with twice as much reagent in addition reactions