



(3) HST Pedal Position



Checking and Adjusting HST Neutral Position

⚠ CAUTION

- Park the tractor on a flat place and keep all the levers at neutral position.
 - Jack up the engine tractor and bring it in the 2WD mode.
1. Fit the HST mechanism first and then exterior components until the engine can get started.
 2. Remove the hex. socket head plug from F and R ports.
 3. Assemble adapter C (07916-50371) and thread joint (07916-50341) with the gasket between them.
 4. Install the assembled adapter C and thread joint F and R ports.
 5. Measure the F and R HST charge pressures. Place 5 MPa (50 kgf/cm², 725 psi) gauges on the two spots in the photo.
 6. Lift the entire steps, as shown below, so that the piston can be adjusted.
 7. If any of the servo piston parts has been replaced, readjust the HST neutral position, referring to the dimension A of the servo piston adjusting screw (2).
 8. Start the engine and measure the F and R charge pressures. Now adjust the piston neutral position so that the F port side pressure and the R port side one be the same. (Take the measurement with the engine rpm at MAX.).
 9. Finally lock the piston adjusting nut and fit the exterior components.

Differential pressure F-R	Factory spec.	-0.2 to +0.2 MPa -2 to +2 kgf/cm ² -20 to +20 psi
---------------------------	---------------	--------------------------------------------------------------------

Condition

- Engine speed : Maximum Speed
- Oil Temperature : 45 to 55 °C (113 to 131 °F)

(Reference)

- Dimension A : 19.4 mm (0.764 in.)

- | | |
|---------------------|------------------|
| (1) Cap Nut | F : Forward Side |
| (2) Adjusting Screw | R : Reverse Side |
| (3) Lock Nut | |
| (4) Cover | |
| (5) Servo Piston | |
| (6) Spring | |

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HST Pedal

■ NOTE

- When removing the HST pedal carry out the following adjustment.

1. Adjust the rod length L so that the face of the HST pedal A becomes horizontal.
2. After adjusting the HST pedal perform "Mode K" (Refer to "4. CHECKING AND ADJUSTING" at "9. ELECTRICAL SYSTEM" section).

(Reference)

- Rod length L : 141 mm (5.55 in.)

- | | |
|---------------|-----------------------|
| (1) HST Pedal | A : Face of HST Pedal |
| (2) Rod | |

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web use the solubility curve to answer these questions determine how much potassium nitrate will dissolve in 20 g of water at 40 °C 200 cm³ of saturated lead(II) nitrate solution was prepared at a temperature of 90 °C

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web collect experimental data and create a solubility curve by the end of this lab students should be able to properly use an analytical balance to measure mass set up an experimental work station to measure the solubility of a salt in water as a function of the temperature generate a workbook using google sheets prior knowledge

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web sample answer sugar has a much greater solubility in hot water than cold water so its solubility curve probably resembles the sloping solubility curve of potassium nitrate temperature solubility g 100 ml 10 °C 36 g 100 ml 20 °C 36 g 100 ml 30 °C 37 g 100 ml 40 °C 37 g 100 ml 50 °C 38 g 100 ml 60 °C 38 g 100 ml 70 °C 39 g 100 ml 80

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web 2 days ago study with quizlet and memorize flashcards containing terms like to test the hypothesis above you changed the of the solvent and measured the of a solid therefore the independent variable was and the dependent variable was the results of this lab experiment the hypothesis and more

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web may 29 2018 figure 2 the graph shows the solubility curve of potassium nitrate with corresponding temperatures as the temperature increases so does the solubility of the solution a y 0 2 1 61 b y 0 2 1 61 0 60 2 1 60 61 0 40 2 1 40 61 8 5 g 100 ml 4 6 g 100 ml

11 thermodynamics of solubility chemistry libretexts - Oct 11 2022

web mar 15 2021 this experiment will further develop your understanding of thermodynamics while relating it to two

concepts you already have studied equilibrium and solubility you will determine three thermodynamic values associated with the solubility of potassium nitrate the reaction we will be studying is the dissolution of potassium nitrate in water

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web analysis construct a solubility curve using the mass of solute that will dissolve in 100 g of water at a specified temperature in this lab 5 ml of water was used assume 10 g ml of water take the collected data make it proportional to get the solubility of the KNO_3 graph temperature vs solubility using a best fit curve

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web part i construct a single graph with multiple solubility curves to clearly show the solubility of the four ionic compounds in the data table below the following tips will help you make the best possible graph