Limiting Reactants Worksheet KEY

Answer the following questions, using stoichiometry to convert between moles of one substance and moles of another substance within a reaction and using molar masses to convert between grams of a substance and moles of that substance.

Given the following reaction: $C_3H_{12} + 8 O_2 \rightarrow 5 CO_2 + 6 H_2O$

 If 5.25 grams of pentane (C₁H₁₂) are completely reacted with oxygen (O₂), how many grams of carbon dioxide (CO₂) will be produced?

5.25 g C₃H₁₂ x
$$\frac{1 \, mol \, C_5 \, H_{12}}{72.151 \, g \, C_5 \, H_{12}} = 0.07276 \, mol \, C_5 H_{12}$$

0.07276 mol C₃H₁₂ x $\frac{5 \, mol \, C \, O_2}{1 \, mol \, C_5 \, H_{12}} = 0.3638 \, mol \, CO_2$
0.3638 mol CO₂ x $\frac{44.009 \, g \, CO_2}{1 \, mol \, CO_2} = 16.0 \, g \, CO_2$

 If 9.75 grams of oxygen (O₂) are completely reacted with pentane (C₅H₁₂), how many grams of carbon dioxide (CO₂) will be produced?

9.75 g O₂ x
$$\frac{1 mol O_2}{31.998 g O_2}$$
 = 0.3047 mol O₂
0.3047 mol O₂ x $\frac{5 mol C O_2}{8 mol O_2}$ = 0.1904 mol CO₂
0.1904 mol CO₂ x $\frac{44.009 g CO_2}{1 mol CO_2}$ = 8.38 g CO₂

3. Based on your answers to question #1 and 2, if 5.25 grams of pentane are reacted with 9.75 grams of oxygen, what is the limiting reactant and how many grams of CO₂ actually form?

Limiting reactant: O2

8.38 g CO2 actually form

Limiting Reactants Answer Key

Morris Hein, Susan Arena

Limiting Reactants Answer Key:

Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career

Kaplan SAT Subject Test Chemistry 2015-2016 Kaplan Test Prep, 2015-03-03 Essential strategies practice and review to ace the SAT Subject Test Chemistry Getting into a top college has never been more difficult Students need to distinguish themselves from the crowd and scoring well on a SAT Subject Test gives students a competitive edge Kaplan's SAT Subject Test Chemistry is the most up to date guide on the market with complete coverage of both the content review and strategies students need for success on test day Kaplan's SAT Subject Test Chemistry features A full length diagnostic test Full length practice tests Focused chapter summaries highlights and guizzes Detailed answer explanations Proven score raising strategies End of chapter quizzes Kaplan is serious about raising students scores we guarantee students will get a higher Foundations of College Chemistry Morris Hein, Susan Arena, 2013-01-01 Learning the fundamentals of chemistry score can be a difficult task to undertake for health professionals For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement SuperSimple Chemistry DK,2020-06-09 A fantastic aid for coursework homework and studying for tests this comprehensive guide covers Next Generation Science Standards for grades 6 10 and will have you ready for tests and exams in no time Each topic is fully illustrated to support the information make the facts crystal clear and bring the science to life A

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Mapping College Chemistry Stephen DeMeo,2019-03-01 This text is a chemistry problem solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first year college General Chemistry course The book incorporates a chemistry problem solving plan one that uses an innovative graphic organizer strategy The strategy successfully evaluated with students combines problem solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry

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