

Name: Date:

Student Exploration: Limiting Reactants

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make?

2. Which ingredient limited the number of hot dogs you could make?

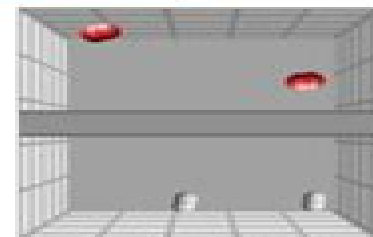
3. Which ingredient will you have leftovers of?

Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.

To begin, make sure $\text{H}_2 + \text{O}_2$ becomes H_2O is selected. The small "2" in H_2 , O_2 , and H_2O is a **subscript**. Subscripts represent the number of atoms in a **molecule**.



1. Use the sliders to set the number of O_2 molecules and H_2 molecules to two.

A. How many hydrogen molecules (H_2) are there?

Hydrogen atoms?

B. How many oxygen molecules (O_2) are there?

Oxygen atoms?

2. How many H_2O molecules do you think will form when these four molecules react?

3. Click **Play** (▶). How many H_2O molecules actually formed?

Limiting Reactant Gizmo Teachers Guide

Rachel Sandford



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Table of Contents Limiting Reactant Gizmo Teachers Guide

1. Understanding the eBook Limiting Reactant Gizmo Teachers Guide
 - The Rise of Digital Reading Limiting Reactant Gizmo Teachers Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Limiting Reactant Gizmo Teachers Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Limiting Reactant Gizmo Teachers Guide
 - User-Friendly Interface

4. Exploring eBook Recommendations from Limiting Reactant Gizmo Teachers Guide
 - Personalized Recommendations
 - Limiting Reactant Gizmo Teachers Guide User Reviews and Ratings
 - Limiting Reactant Gizmo Teachers Guide and Bestseller Lists
5. Accessing Limiting Reactant Gizmo Teachers Guide Free and Paid eBooks
 - Limiting Reactant Gizmo Teachers Guide Public Domain eBooks
 - Limiting Reactant Gizmo Teachers Guide eBook Subscription Services
 - Limiting Reactant Gizmo Teachers Guide Budget-Friendly Options
6. Navigating Limiting Reactant Gizmo Teachers Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Limiting Reactant Gizmo Teachers Guide Compatibility with Devices
 - Limiting Reactant Gizmo Teachers Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Limiting Reactant Gizmo Teachers Guide
 - Highlighting and Note-Taking Limiting Reactant Gizmo Teachers Guide
 - Interactive Elements Limiting Reactant Gizmo Teachers Guide
8. Staying Engaged with Limiting Reactant Gizmo Teachers Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Limiting Reactant Gizmo Teachers Guide
9. Balancing eBooks and Physical Books Limiting Reactant Gizmo Teachers Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Limiting Reactant Gizmo Teachers Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Limiting Reactant Gizmo Teachers Guide
 - Setting Reading Goals Limiting Reactant Gizmo Teachers Guide
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Limiting Reactant Gizmo Teachers Guide
 - Fact-Checking eBook Content of Limiting Reactant Gizmo Teachers Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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