



# **GOUT** and its **CURE**

**J. Compton Burnett, M.D.**



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arrhenius's definition of acids and bases the earliest definition of acids and bases is arrhenius's definition which states that an acid is a substance that forms hydrogen ions  $H^+$  when dissolved in water and a base is a substance that forms hydroxide ions  $OH^-$  when dissolved in water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other's chemistry to produce a rather innocuous substance water in fact the

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