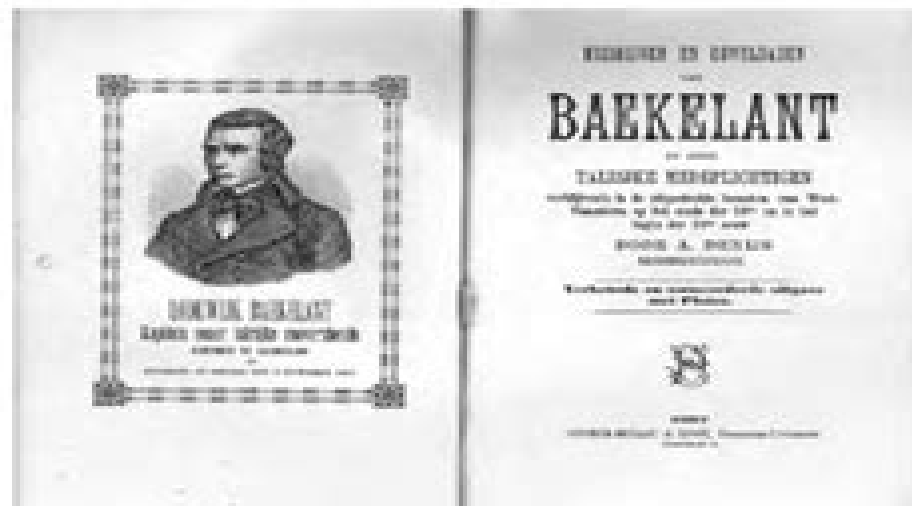


In al dan niet "officiële" vertalingen waren hier bijzonder populair: Bulwer-Lytton en zijn *Pompeji*, (The last days of Pompeii), Beecher-Stowe (De hut van Oom Tom), K. de Montépin, Dumas, Féval, d'Ennery, Souvestre en vooral Eugène Sue wiens *Wandelende Jood* en *De mysteriën van het Volk* (later *De Verborgenheden des Volks*) overvloedig worden vertaald en gedrukt.

Maar er is ook de eigen productie met Henri Conscience (1812-1883) op kop en die Conscience komt zelfs in Frankrijk veelvuldig aan de hak n.a. in *La Lecture, Journal du Jeudi, L'Ouvrier*.

Bijzonder veel gelezen is ook Pieter Geiregat (1828-1902) met zijn vele historische werken, een genre dat ook P. Cautereels (1833-1903) beoefent. Uiteraard is Conscience hen daarin te sterk, hij die overvuldig romantisme aan historie koppelt. Zeer geliefd wordt Pieter Ecrivisse met *De Treuen* (1844), *De helderijders* (1845) en *De verdronken* (1859). En er is ook Emiel Dobbelaere met *De Geheimen van Gine* (na die van Parijs door Sue, die van Londen door Féval en in afwachting van die van Antwerpen, Brussel en Amsterdam). Ook "historische criminelen" vormen een geliefd onderwerp, die het via volksverlevering en volkslied tot vaak anonieme volksroman brengen zoals *Baekelant en zijnne cultijle medeplichtigen* of nog *Het wonderbaar leven van Cartouche*, beiden te Gent door Snoeck-Ducayx uitgegeven.

En dit soort boeken kent - gezien het succes - ontelbare plagiatissen, na-apers en kopieerders.



Baekelant en zijnne cultijle medeplichtigen, Snoeck - Ducayx Gent

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Michal Rosen-Zvi



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oc the mixtures are stirred and then allowed to sit for three hours before measuring the amount of solid that dissolves ten

separate trials are conducted for each experiment

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on entropy and interparticle forces another factor affecting solution formation is the enthalpy of the solution or the energy

required for the solution to form

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web if a solute is soluble in a solvent the overall process is said to be spontaneous if a solute does not dissolve in a solvent insoluble the solution process is said to be non spontaneous a spontaneous process or change proceeds in a given direction without needing to be driven by an external energy input

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web solubility and solubility product mh 5 16 1 16 2 in this section we are going to consider the solubility of ionic solids in water solubility may be considered to be an equilibrium the equilibrium is between solid and ions in solution any ionic solid is 100% ionized in aqueous solution once it actually dissolves

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web pogil solubility why solubility is a measure of the amount of solute that will dissolve in a given amount of solvent usually

water at a given temperature a solubility curve shows how much solute dissolves in a given volume of solvent at a given temperature

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web information solubility rules a solute is considered soluble if an appreciable amount of it can be dissolved in a given amount of the solvent for example both table salt NaCl and table sugar $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ are soluble substances in water a solute is considered insoluble if very little of it dissolves in a given amount of the solvent

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web predict what might happen to the solubility of the insoluble salt if a 1.0 M acid solution was added drop wise to the beaker hint which ion species in the beaker will react with the acid how will this reaction alter the concentration of the common ion