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- 1. Learnors do TWO experiments (ONE Chemistry, ONE Physics) for SSA. Choose three experiments from this ATP.
- 2. Term 1: Choose ONE experiment. Record its term one.
- 3. Term 2 OR Term 3: Choose ONE experiment. Record the mark in term 2.
- 4. The suggested Termal experiments NOT chosen for SSA, should become informal experiments.

[illegible]

TERM 1 (52 days)	Week 1 15/1 – 17/1 (3 days)	Week 2 20/1 – 24/1 (5 days)	Week 3 27/1 – 31/1 (5 days)	Week 4 3/2 – 7/2 (5 days)	Week 5 10/2 – 14/2 (5 days)	Week 6 17/2 – 21/2 (5 days)	Week 7 24/2 – 28/2 (5 days)	Week 8 3/3 – 7/3 (5 days)	Week 9-10 10/3 – 21/3 (9 days)	Week 11 24/3 – 28/3 (5 days)
CAPS Topics	MATTER & MOTION 1: Kinematics 1: Inquire (2 hrs)	MATTER & MOTION 1: Kinematics 2: Inquire (2 hrs)	MATTER & MOTION 1: Kinematics 3: Vertical projectile motion (2 hrs)	MATTER & MOTION 1: Kinematics 4: Vertical projectile motion (2 hrs)	MATTER, SOUND & LIGHT: Doppler Effect (2 hrs)	MATTER & MATERIALS: Organic molecules (2 hrs)	MATTER & MATERIALS: Organic molecules (2 hrs)	MATTER & MATERIALS: Organic molecules (2 hrs)	MATTER & MATERIALS: Organic molecules (2 hrs)	COTWOL TEST (2 hrs) Discussion and remedial work on test (2hrs)
Topics / Concepts, Skills and Values	- Define & calculate the momentum of a moving object $p = mv$ - Describe the vector nature of momentum & draw vector diagrams - State Newton's second law in terms of momentum $F = \frac{\Delta p}{\Delta t}$ - Calculate the change in momentum when a constant force acts on an object.	- Define Inquire - Use the impulse-momentum theorem ($F \Delta t = \Delta p$) in calculations for a variety of situations (one dimension) - Inquire and safety considerations - State the principle of conservation of linear momentum - Explore what is meant by an isolated system, internal and external forces - Apply conservation of momentum to collisions of two objects (one dimension) - Distinguish between elastic and inelastic collisions by calculation	- Explore what is meant by a projectile - Use equations of motion to determine the position, velocity and displacement of a projectile at any given time - Sketch v vs t and a vs t graphs for a free falling object, an object thrown vertically upwards, object thrown vertically downwards & launching objects - Use equations for position versus time and velocity versus time for the graphs of 1D projectile motion	- For given v vs t or a vs t graphs, determine position, displacement and velocity or acceleration at any time t - For given v vs t or a vs t graphs, describe the motion of an object thrown vertically upwards, thrown vertically upwards & thrown vertically downwards and so on	- State the Doppler effect and explain (using illustrations) the change in pitch observed when a source moves toward or away from a listener (sound) and ultrasonics - State Applications of the Doppler effect. (Example: ultrasound in medicine) - State problems using $f_L = \frac{v \pm v_L}{v \pm v_S} f_s$ when $v > 0$ is towards source or listener moves, state sign, explain red shift & use the Doppler Effect to explain why we conclude that the universe is expanding.	- Define Inquire, molecules, functional group, hydrocarbon, homologous series, saturated, unsaturated, and structural isomer - Write condensed, structural & molecular formulae (max 8 C atoms, 1 functional group per molecule) for alkanes (no rings), alkenes (no rings), alcohols, halo alkanes (no rings), carboxylic acids, aldehydes, ketones, esters - Write IUPAC names for structural condensed structural formulae for compounds, both above series	- Write IUPAC names from structural or condensed structural formulae for compounds (max 8 C atoms, 1 functional group per molecule) - Identify alkyl substituents (methyl and ethyl) & max limited alkyl substituents - Identify compounds that are saturated, unsaturated, structural isomers (chain, positional, functional), Physical properties: boiling point, melting point, vapour pressure	- Relationship between physical properties and strength of intermolecular forces, chain length and branching - Combination of alkanes in excess oxygen and use as fuels - Equation & reaction conditions for the formation of an ester and IUPAC names for reactant and products - Classify reactions as elimination, addition or substitution. Equations and reaction conditions for addition reactions of alkenes	- Equations and reaction conditions for elimination reactions: dehydrohalogenation of halo alkanes, cracking of alkanes, dehydration of alcohols - Equations and reaction conditions for substitution reactions: hydrolysis of halo alkanes, hydrogenation of alkenes	Revision (2 hours) END PAPER (100 marks/ 2 hours), 60 marks + Physics - Newton's laws of motion - Momentum and impulse - Vertical projectile motion - Doppler Effect 40 marks + Chemistry - Organic molecules Discussion and remedial work of content test.
Requisite pre-knowledge	Newton's laws of motion	Newton's laws of motion Equations of motion	Equations of motion	Equations of motion	Wave properties: frequency, wavelength, amplitude Graphical representation in waves	Chemical bonding: Valency	Chemical bonding: Valency	Thermodynamic laws	IUPAC naming: writing different formulae	N/A

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