

GRADE 9 NATURAL SCIENCES JUNE TEST

SECT						
- 6	0.000	s options are provided as answers rect answer and write CNLY the L ION NUMBER (1.1.1-1.1.10) on y	ETTER	(A-D) next to the		
1,1,1	Which of the following can you NOT learn from a chemical formula?					
	A	Elements in the compound	8	Phase of the element		
	C	Ratio of atoms	D	Total number of atoms		
1.1.2	Which of the following is NOT a sub-atomic particle?					
	A	Electrons	. 8	Protons		
	0	Neutrons	0	Nucleons		
1.13	What is the prefix for 6 stores?					
	A	Hepta-	8	Hexa-		
	C	Pero-	0	Tri-		
1.1.4	An element has an atomic number of 30 and mass number of 65. How many nucleons will if have?					
	A	30	8	35		
	C	65	0	95		
1.1.5	Element at group 15, period 2 on the periodic table					
	A	Carbon	8	Nitrogen		
	100	Silicon	. 0	Oxygen		

	180	Silicon		Lithium	
	C	Magnesium	0	Carbon	
17	Subs	stances on the left side of a chemi	cal equa	flort:	
	A	Products		Reactants	
	C.	Metals	0	Non-metals	
33	Which	it of the following is NOT a way to	preven	t rust?	
	A	Galvanisation		Electrolysis	
	¢	Paint	0	Ollor grease	
19	Sut	stance with a pH lower than 7 wil	ibe dan	sified as:	
	A	Lemon Juloe		Neutral	
	¢	Apde	0	Akaine	
3.30	An	neutral substance would turn	with	universal indicator.	
	A	Green		Volet.	
	ė.	Colourless		Red	[Mat]
		e the correct TERM for the follow			
		e the correct TERM for the follow next to the QUESTION NUMBER			
T	ERM		b(1.2.1-	(2.7) on your answer	
	ERM Mea	need to the QUESTION NUMBER	b(1.2.1-	(2.7) on your answer	
21 22	ERM Mea Reac	need to the QUESTION NUMBER sure of how strongly acidic or also	t (1.2.1- line a su	(2.7) on your answer	
21 22 22 23	ERM Mea Rea Nam	next to the QUESTION NUMBER sure of how strongly acidic or alka stion of a substance with oxygen.	t (1.2.1- line a su lie.	(27) on your answer	
21 22 23 24	ERM Mea Read Nam Proc	next to the QUESTION NUMBER sure of how strongly acidic or also stion of a substance with oxygen. e for group 18 on the periodic tab	t (1.2.1- line a su lie.	(27) on your answer	
2! 22 23 24 25	ERM Mea Reac Nam Proo H ₂ (next to the QUESTION NUMBER sure of how strongly acidic or alka stion of a substance with oxygen. e for group 18 on the periodic tab ess whereby compounds can be i	t (1.2.1-) Vine a su Vie Inchen d	(2.7) on your answer to be a second of the s	

1.1.6 The following is an example of a semi-metal:

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- 1. Understanding the eBook Grade 9 Natural Science June Exam Papers
 - The Rise of Digital Reading Grade 9 Natural Science June Exam Papers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Grade 9 Natural Science June Exam Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 9 Natural Science June Exam Papers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Grade 9 Natural Science June Exam Papers
 - Personalized Recommendations
 - Grade 9 Natural Science June Exam Papers User Reviews and Ratings
 - Grade 9 Natural Science June Exam Papers and Bestseller Lists
- 5. Accessing Grade 9 Natural Science June Exam Papers Free and Paid eBooks
 - Grade 9 Natural Science June Exam Papers Public Domain eBooks
 - Grade 9 Natural Science June Exam Papers eBook Subscription Services
 - Grade 9 Natural Science June Exam Papers Budget-Friendly Options
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 - Participating in Virtual Book Clubs
 - \circ Following Authors and Publishers Grade 9 Natural Science June Exam Papers
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 9 Natural Science June Exam Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 9 Natural Science June Exam Papers
 - Setting Reading Goals Grade 9 Natural Science June Exam Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 9 Natural Science June Exam Papers
 - Fact-Checking eBook Content of Grade 9 Natural Science June Exam Papers
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 - Interactive and Gamified eBooks

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