

1

11

[3]

F51

161

[3]

The fundamental ideas in chemistry

- electron
 nucleus
 neutron
 each for I mark
- 2. (a) (i) B (ii) D
 - (b) A and B only one type of atom

 C and D more than one type of atom

 accept element for atom
 ignore the word 'mixture'

 (chemically) bonded

accept (chemically) joined or similar idea of joined

- 3. B carbon monoxide 1
 CO accept carbon oxide 1
 - do **not** credit carbon dioxide do **not** credit if any superscripts **or** subscripts used but accept C_1O_2 , accept OCdo **not** credit if obviously lower case
 - G water
 H₂O accept hydrogen oxide
 do not accept hydrogen hydroxide
 do not credit if obviously lower case or if 2 not subscript
 do not accept HOH
 accept OH₂
 - D ammonia do not accept ammonium 1
 NH₃ do not credit if obviously lower case, or if 3 not subscript 1
 accept nitrogen hydrogen nitrate or nitrite
 allow H₃N
- 4. correct reactants (i.e. sodium + water either way round)
 - correct products (i.e. sodium hydroxide + hydrogen, either way round)
 - acrow → / =

fdo <u>not</u> allow produce/makes or similar] fdo <u>not</u> allow symbols or formulae] each for 1 mark

correct use of 'react//reaction'/reactants'/combine (not mixed/added/join) 3
correct use of 'produces'/products'/gives/forms/makes/creates reactants
correctly identified

each for 1 mark

products correctly identified



1

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Christian Drosten

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