

Final Exam Study Guide (General Chemistry I)

- Chemistry - the study of matter and the changes that matter undergoes
- Matter - anything that has mass and takes up space
 - Three states of matter: solid, liquid, and gas (plasma is a gas)

	Proximity of Particles	Energetics	Size/Shape
Solid	Closely packed and in direct contact with each other	Have very little energy of motion (slow)	Fixed shape and fixed volume
Liquid	In indirect contact with each other; usually about a particles distance apart (bounce off each other)	Have an intermediate amount energy of motion (medium)	Does not have a fixed shape; has a fixed volume
Gas	Particles are very far apart	Have a large amount of energy of motion (fast)	No fixed shape or volume

- **PARTICLES ARE ALWAYS MOVING!**
- "Can you undo it?" (Page 12)
 - Physical Changes - changes physical appearance but not its composition
 - Ex. Melting, evaporation, freezing, etc.
 - Chemical Changes - changes in chemical composition of matter (chemical reaction)
 - Ex. Baking soda and vinegar, burning, mixing, etc.
 - Physical Properties - characteristics that can be described by observing the matter; no change in composition is required to detect a physical property
 - Ex. Color, odor, density, etc.
 - Chemical Properties - a change in composition is required to detect a chemical property
 - Ex. Flammability
- Accuracy Versus Precision:
 - In every measurement, there is some degree of uncertainty.
 - **NUMBERS WITHOUT UNITS ARE MEANINGLESS IN MEASUREMENT!**
 - Accuracy - the agreement of a measurement with a known or accepted value (correctness)
 - Precision - the agreement of a measurement with other measurements in a given set of measurements (reproducibility/repeatability)
 - You can have accuracy without precision, and precision without accuracy.
 - **WE ARE ONLY AS CERTAIN AS OUR LEAST CERTAIN NUMBER!**
 - 254 can mean 253.5 to 254.4
 - 254.0 can mean 253.95 to 254.04
 - 254.00 can mean 253.950 to 254.004
 - 254.000 can mean 253.9500 to 254.0004
- Density - the mathematical relationship of mass to volume

Grade 12 Chemistry Final Exam Study Guide

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learning In this approach simulations and games could play a significant role by addressing many goals and mechanisms for learning science the motivation to learn science conceptual understanding science process skills understanding of the nature of science scientific discourse and argumentation and identification with science and science learning To explore this potential Learning Science Computer Games Simulations and Education reviews the available research on learning science through interaction with digital simulations and games It considers the potential of digital games and simulations to contribute to learning science in schools in informal out of school settings and everyday life The book also identifies the areas in which more research and research based development is needed to fully capitalize on this potential Learning Science will guide academic researchers developers publishers and entrepreneurs from the digital simulation and gaming community and education practitioners and policy makers toward the formation of research and development partnerships that will facilitate rich intellectual collaboration Industry government agencies and foundations will play a significant role through start up and ongoing support to ensure that digital games and simulations will not only excite and entertain but also motivate and educate

Newly Hired Teachers of Science Julie A. Luft, Shannon L. Dubois, 2015-12-09 Supporting newly hired science teachers has taken on an increased importance in our schools This book shares the most current information about the status of newly hired science teachers different ways in which to support newly hired science teachers and different research approaches that can provide new information about this group of teachers Chapters in the book are written by those who study the status of beginning science teachers mentor new teachers develop induction programs and research the development of new science teachers Newly Hired Teachers of Science is for administrators who have new science teachers in their schools and districts professionals who create science teacher induction programs mentors who work closely with new science teachers educational researchers interested in studying new science teachers and even new science teachers This is a comprehensive discussion about new science teachers that will be a guiding document for years to come Final Exam Review: Elementary Algebra A. A. Frempong, Elementary Algebra covers Signed Number and Real Number Operations Order of Operations and Evaluation of Expressions Exponential Notation and Rules of Exponents Polynomial addition subtraction multiplication and division Solving First Degree Equations Word Problems Ratio and Proportion Factoring Polynomials Solving quadratic equations by factoring Graphs Slopes Intercepts and Equations of Straight Lines Solving Systems of Linear Equations and Word Problems Radicals square roots addition Pythagorean Theorem and Applications Areas and Perimeters Algebraic Fractions reduction multiplication division Solving Linear inequalities **Review Guide for LPN-LVN Pre-entrance Exam** Mary McDonald, 2004 These best selling review guides provide an overview of the math science and verbal content necessary for admission to AD BS LPN and LVN programs in nursing Each include approximately 1 000 sample questions and three practice exams in the areas of math science and verbal and contain helpful tips for test preparation

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