#### AP Biology Study Guide Unit 1

#### Properties of Water:

- 1. Polarity allows H20 to form Hydrogen Bonds
- 2. Cohesion Water H bonds to itself, creating surface tension
- 3. Adhesion Water adheres to other polar substances
  - a. Allows easy transport of water against Gravity
- 4. Universal Solvent Water can dissolve Polar and Ionic Compounds
  - a. This allows for reactions between Aqueous Solutions
- 5. Expands when Frozen this means that Ice floats
- High Specific Heat water can absorb or release a large amount of heat with only a slight change in its own temperature (caused by cohesion)
  - High Heat of Vaporization only warmest molecules can evaporate, so this causes evaporative cooling
- 7. Can be regulated by non-polar hydrophobic substances
- Neutral PH → can form acids or bases if substances are added

Polarity - Difference in electronegativity between atoms bonded together create slightly charged ends of the molecules they form

Hydrophobic Substance - Nonpolar substance that repels water Hydrophilic Substance - Polar substance that attracts water molecules

#### CHON - Carbon, Hydrogen, Nitrogen, Oxygen

- Makeup 99% of living matter
- Strong Biological Cornerstones Because:
  - Smallest Diameter/Valency for atoms of 1, 2, 3, 4 valence electrons
  - Form Stable Covalent Bonds
  - Can form double and triple bonds
  - Versatile

#### Biological Transition Metals - Mn. Fe. Co. Cu. Zn.

Can exist in multiple oxidation states → can be both oxidized and reduced

#### Biological Ions: Na+, Mg2+, Cl-, K+, Ka2+

Used to maintain homeostasis and osmotic concentration

Phosphorous/Sulphur - Energy Carriers which form strong high energy bonds

#### 5 Foundation Biological Molecules:

# **Guide Line Study For Ap Biology Campbell**

Joel Rodriguez-Saldana

#### **Guide Line Study For Ap Biology Campbell:**

The Million Person Study of Low-Dose Radiation Health Effects John D. Boice Jr, André Bouville, Lawrence T. Dauer, Ashley P. Golden, Richard Wakeford, 2024-11-25 This book presents original research findings of The Million Person Study of Low Dose Radiation Health Effects MPS the largest and most comprehensive epidemiologic study of its kind to investigate the health effects of low level chronic radiation exposure on American workers and veterans throughout the 20th century Since the early 1900s epidemiologists have studied the consequences of radiation exposures yet the health effects of low levels received gradually over time remain unresolved This uncertainty comes at a time when the public and workers are experiencing ever increasing levels of radiation exposure from advances in medical radiation imaging techniques e g CT scans frequent flying at high altitudes and environmental and occupational exposures The MPS is providing answers by studying 30 radiation exposed U S populations including workers at nuclear power plants radiologists workers at former Manhattan Project sites nuclear submariners nuclear weapons test participants atomic veterans industrial radiographers and radium dial painters Ongoing for more than 20 years and coordinated by the National Council on Radiation Protection and Measurements Vanderbilt University Medical Center and Memorial Sloan Kettering Cancer Center the MPS is a national effort supported by the Department of Energy National Aeronautics and Space Administration U S Navy Defense Threat Reduction Agency Nuclear Regulatory Commission Centers for Disease Control and Prevention and the Environmental Protection Agency Unparalleled in scope and quality the MPS provides an understanding of low dose health effects that is desperately needed for decision makers and the radiation protection community as society continues to increase the uses of radiation technologies Individual chapters were originally published in the International Journal of Radiation Biology

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