

ANSWER GUIDE PHYSICAL SCIENCES: GRADE 12 CLASS TEST: ACIDS & BASES

Written on: 25.06.2020

Total Marks: 65

1	1.1	B / / C / /	24.00
2	Hyd	rechleric acid is a strong acid and sedium hydroxide is a strong base, therefore the equivalence point be reached at a pH close to 7, which is in the colour change range of bromothymol blue.	2
3	3.1	An acid is a proton donor.	1
	3.2	A concentrated acid is an acid solution with a high proportion of acid (solute) to water (solvent), 🗸 🗸 regardless of the degree of ionisation or dissociation of the acid in the solution. A strong acid is an acid that dissociates or ionises almost completely when in solution, regardless of its concentration. 🗸 🗸	2
4	4.1 4.2 4.3 4.4 4.5	LIOH // HJPO, // HNO, // HOOOH // NaHCO, //	
5 6	H ₂ S	O ₄ + 2H ₂ O ✓ → 2H ₃ O ✓ + SO ₄ ² ✓ OR: H ₂ SO ₄ + H ₂ O → H ₃ O + HSO ₄ · A standard solution is a solution of known concentration. ✓ ✓	
	6.2	c = m/MV m = cMV / = (0,15){106)(0,25) / = 3,975 g Na ₂ CO ₃ /	
	6.3	$\frac{n_a}{n_b} = \frac{c_a V_a}{c_b V_b} \therefore c_a = \frac{n_a c_b V_b}{n_b V_a} \checkmark$ $= \frac{(1)(0.15)(25) \checkmark}{(1)(27.5) \checkmark}$ $= 1.38 \times 10^{-1} \text{ mol.dm}^{-3} \checkmark$	
7	рН	= -log[H *] / and: HCl dissociates completely 0,2 mol.dm -3 H * ions = -log[0,2] / = 0,68 /	2

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Catherine M. Jackson

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Morse, 1909 Goyal's ICSE Chemistry Question Bank with Model Test Papers Class 10 for 2026 Examination
, 2025-07-02 Goyal's I.C.S.E. Chemistry with Model Test Papers Class 10 for 2023 Examination GBP Editorial, 2022-08-10
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