

2 homework point(s)

## Galvanic Cells

Using tabulated standard reduction potentials from your text, calculate the standard cell potential,  $E^\circ_{\text{cell}}$  (always positive for a galvanic cell), based on the following (unbalanced) reaction:



$$E^\circ_{\text{cell}} = 0.036 \text{ V}$$

of Table A5.5, p. A24 Zumdahl "Chemical Principles" 7th ed.

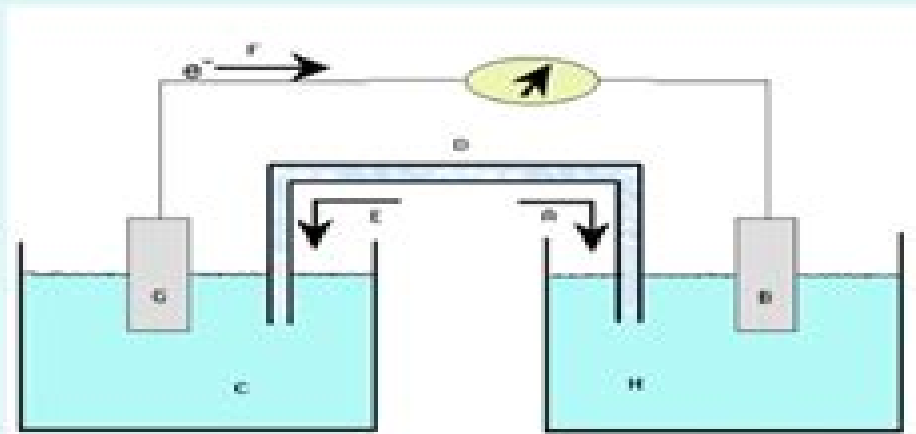
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A galvanic cell based on the above reaction is constructed according to the generic sketch of a galvanic cell shown below:



Choose the label which best describes the following:

C: Solution containing  $\text{Cr}^{3+}$ ,  $\text{H}^+$  and  $\text{Cr}_2\text{O}_7^{2-}$

B:  $\text{Cl}_2(\text{g})/\text{Pt}$  cathode

The overall, balanced, spontaneous reaction occurring in this (standard) cell is:



False: The salt bridge disrupts the electrical circuit while permitting mixing of the solutions in the anode and cathode compartments.

False: The salt bridge is labeled "D".

True: The solution in tube "D" must contain an electrolyte for the cell to operate.

True: The salt bridge is intended to allow counter ions to flow between the two cell compartments.

False: Tube "D" allows  $\text{Cr}^{3+}$  ions to migrate readily to the other compartment and to react directly with the chlorine gas.

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# Lon Capa Solution Guide Phys

**David Halliday, Robert Resnick, Jearl Walker**



## **Lon Capa Solution Guide Phys:**

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**FUNDAMENTALS OF PHYSICS EXTENDED** **Fundamentals of Physics, Extended** David Halliday, Robert Resnick, Jearl Walker, 2013-08-13 The 10th edition of Halliday's Fundamentals of Physics Extended building upon previous issues by offering several new features and additions The new edition offers most accurate extensive and varied set of assessment questions of any course management program in addition to all questions including some form of question assistance including answer specific feedback to facilitate success The text also offers multimedia presentations videos and animations of much of the material that provide an alternative pathway through the material for those who struggle with reading scientific exposition Furthermore the book includes math review content in both a self study module for more in depth review and also in just in time math videos for a quick refresher on a specific topic The Halliday content is widely accepted as clear correct and complete The end of chapters problems are without peer The new design which was introduced in 9e continues with 10e making this new edition of Halliday the most accessible and reader friendly book on the market WileyPLUS sold separately from text

**Physics for Scientists and Engineers with Modern Physics** Douglas C. Giancoli, 2008 Key Message This book aims to explain physics in a readable and interesting manner that is accessible and clear and to teach readers by anticipating their needs and difficulties without oversimplifying Physics is a description of reality and thus each topic begins with concrete observations and experiences that readers can directly relate to We then move on to the generalizations and more formal treatment of the topic Not only does this make the material more interesting and easier to understand but it is closer to the way physics is actually practiced Key Topics INTRODUCTION MEASUREMENT ESTIMATING DESCRIBING MOTION KINEMATICS IN ONE DIMENSION KINEMATICS IN TWO OR THREE DIMENSIONS VECTORS DYNAMICS NEWTON'S LAWS OF MOTION USING NEWTON'S LAWS FRICTION CIRCULAR MOTION DRAG FORCES GRAVITATION AND NEWTON'S 6 SYNTHESIS WORK AND ENERGY CONSERVATION OF ENERGY LINEAR MOMENTUM ROTATIONAL MOTION ANGULAR MOMENTUM GENERAL ROTATION STATIC EQUILIBRIUM ELASTICITY AND FRACTURE FLUIDS OSCILLATIONS WAVE MOTION SOUND TEMPERATURE THERMAL EXPANSION AND THE IDEAL GAS LAW KINETIC THEORY OF GASES HEAT AND THE FIRST LAW OF THERMODYNAMICS SECOND LAW OF

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