

Macromolecules - Answers

1. An amino acid contains a central carbon atom, a hydrogen atom, a carboxyl group, an amino group group, and a differing R group.
2. What is a dipeptide? What type of bond is involved? = two amino acids joined via a peptide bond.
3. A nucleotide is a monomer of nucleic acids. There are 4 (how many) different types of nucleotides. (for either RNA or DNA)
4. State two functions of fats.
long term energy storage
insulation
lubrication
cushioning
5. What is the structure of a phospholipid? hydrophilic head group, with a phosphate - always charged and a group of atoms (ex. choline, serine); two hydrophobic tails, fatty acids with long hydrocarbon chains
6. ATP is a molecule that stores a great deal of energy within its phosphate bonds. True or False? True
7. Name and state the structure of a common dissaccharide.
two sugars joined via covalent bond; ex. glucose-glucose = maltose; glucose-fructose = sucrose
8. Name and state the structure of a common polysaccharide.
glycogen and starch = polymers of alpha glucose; storage form of glucose in animals and plants
cellulose = polymers of beta glucose; rigid structure in plant cell walls
chitin = polymer of glucosamine; very strong and flexible; found in exoskeletons of arthropods
9. Heating an enzyme to a certain temperature usually destroys its enzymatic activity even though the same sequence of amino acids remains. Name and explain the levels of structure that have been affected by heating.
secondary structure = alpha helices, beta pleated sheets, random coils
tertiary structure = overall three dimensional conformation, or shape, of the polypeptide
quaternary structure = assembly of subunits

primary structure = chain of amino acids; usually remains intact upon denaturation due to strong peptide bonds.

Macromolecules Guide Answers

KJ Lindholm-Leary



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