

# User's Guide

## *Huskylock* 325



# Manual For Huskylock 36

**C Cleary**



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web for example take the reaction of lead ii nitrate with hydrochloric acid to form lead ii chloride and nitric acid shown below  
$$\text{Pb(NO}_3)_2 \text{ (aq)} + 2 \text{HCl (aq)} \rightarrow \text{PbCl}_2 \text{ (s)} + 2 \text{HNO}_3$$

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web answer key to practice problems on net ionic equations molecular  $\text{AgNO}_3 \text{ (aq)}$   $\text{KCl (aq)}$   $\text{AgCl (s)}$   $\text{KNO}_3 \text{ (aq)}$  total ionic  $\text{Ag}^+ \text{ (aq)}$   $\text{NO}_3^- \text{ (aq)}$   $\text{K}^+ \text{ (aq)}$   $\text{Cl}^- \text{ (aq)}$   $\text{AgCl (s)}$

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web you may have to consult the solubility rules write the net ionic equation for the reaction of  $\text{BaCl}_2(\text{aq})$  and  $\text{Na}_2\text{SO}_4(\text{aq})$  you may have to consult the solubility rules write

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