

Reviewing Content

42. The solvent is the substance in which the solute is dissolved.
43. Random collisions of the solvent molecules with the solute particles provide enough force to overcome gravity.
44. **solubility**: the amount of a substance that dissolves in a given quantity of solvent at specified conditions of temperature and pressure to produce a saturated solution.
saturated solution: a solution containing the maximum amount of solute for a given amount of solvent at a constant temperature and pressure. **unsaturated solution**: a solution that contains less solute than a saturated solution at a given temperature and pressure. **miscible**: describes liquids that dissolve in each other. **immiscible**: describes liquids that are insoluble in each other.
45. Particles of solute crystallize.
46. No; if there were undissolved solute, the excess solute would come out of a supersaturated solution.
47. 5.55×10^2 g AgNO_3
48. Solubility increases with pressure.
49. a. 1.6×10^{-2} g/L
b. 4.7×10^{-2} g/L
50. Dilute and concentrated are relative terms and are not quantitative. Molarity provides the exact number of moles of solute per liter of solution.
51. Molarity is the number of moles of solute dissolved in one liter of solution.
a. 1.3M KCl
b. 3.3×10^{-1} M MgCl_2
52. 2.00×10^1 mL
53. a. 5.0×10^{-1} mol NaCl, 29 g NaCl
b. 1.0 mol KNO_3 , 1.0×10^2 g KNO_3
c. 2.5×10^1 mol CaCl_2 , 2.8 g CaCl_2
54. a. 2.3×10^1 g NaCl
b. 2.0 g MgCl_2
55. a. 16% (v/v) ethanol
b. 63.6% (v/v) isopropyl alcohol
56. Colligative properties are properties of a solution that depend only on the number of solute particles; boiling-point elevation, freezing-point depression, and vapor-pressure lowering. Boiling points are elevated because shells of solvent form around solute particles, reducing the amount of solvent molecules that have sufficient energy to escape the solution; relative to the pure solvent, the amount of energy required to cause vaporization or boiling increases. Solutes disrupt the ordering of the solvent structure, so more kinetic energy must be withdrawn from a solution for it to solidify. This lowers the freezing point of the solution.
57. a. sea water
b. 1.5M KNO_3
c. 0.100M MgCl_2
58. The effective molality of the $\text{Ca}(\text{NO}_3)_2$ solution is 3m. The effective molality of the NaNO_3 solution is 2m.
59. When vapor pressure is lowered relative to pure solvent, more energy must be supplied to reach the boiling point; thus the boiling point is increased relative to pure solvent.
60. The salt lowers the freezing point of the ice-water cooling mixture.
61. 1M solution: 1 mol of solute in 1 L of solution; 1m solution: 1 mol of solute in 1000 g of solvent
62. Add 27.0 g H_2O to 32.0 g CH_3OH .
63. a. 100.26°C
b. 101.54°C
64. a. -4.46°C
b. -2.2°C
65. a. -1.1°C
b. -0.74°C
c. -1.5°C

Understanding Concepts

66. a. The freezing-point depression is twice as great for solute B; solute B must provide twice as many particles in solution.

Mastering Chemistry Answer Key Chapter 5

Brendan G. Carr



Mastering Chemistry Answer Key Chapter 5:

High School Chemistry Unlocked The Princeton Review, 2016-11-29 UNLOCK THE SECRETS OF CHEMISTRY with THE PRINCETON REVIEW High School Chemistry Unlocked focuses on giving you a wide range of key lessons to help increase your understanding of chemistry With this book you ll move from foundational concepts to complicated real world applications building confidence as your skills improve End of chapter drills will help test your comprehension of each facet of chemistry from atoms to alpha radiation Don t feel locked out Everything You Need to Know About Chemistry Complex concepts explained in straightforward ways Walk throughs of sample problems for all topics Clear goals and self assessments to help you pinpoint areas for further review Guided examples of how to solve problems for common subjects Practice Your Way to Excellence 165 hands on practice questions seeded throughout the chapters and online Complete answer explanations to boost understanding Bonus online questions similar to those you ll find on the AP Chemistry Exam and the SAT Chemistry Subject Test High School Chemistry Unlocked covers Building blocks of matter Physical behavior of matter Chemical bonding Chemical reactions Stoichiometry Solutions Acids and bases Equilibrium Organic chemistry Radioactivity and more

Mastering Chemistry Jason Chin, Vault Reports (Firm), Job Vault, 1998 The periodic table never looked so good From the properties of matter and mole calculations to the kinetics of reactions and enthalpy change this guide demystifies the complexities of chemistry through easy to follow charts graphs and sample problems

Mastering Medical Coding - E-Book Marsha Diamond, 2006-06-02 Expansion of ICD 9 CM information Sample patient charts include explanatory notes A simulated medical practice identified as Godfrey Regional lets you study in a real world scenario Key Terms lists highlight the most important vocabulary and content More exercises

CBSE Class 12 Mastering Organic Reactions MCQ With Answers Priti Singhal, 2024-11-11 The primary objective of this book is to serve as a comprehensive guide for students educators and researchers by focusing on reaction mechanisms practical applications and problem solving techniques Organic chemistry is not just about memorizing equations and formulas it is about understanding how molecules interact change and influence each other under different conditions With that in mind this book emphasizes the logic and patterns behind organic reactions making it easier for readers to apply concepts across a variety of scenarios Each chapter of this book builds upon foundational knowledge ensuring a progressive learning experience From nucleophilic substitutions to pericyclic reactions and from oxidation reduction mechanisms to named reactions we cover both fundamental and advanced topics to cater to students at all levels Real world examples have been integrated throughout the chapters to show how organic reactions play essential roles in pharmaceuticals biochemistry agriculture and environmental science This approach bridges the gap between theory and practical applications helping readers appreciate the relevance of organic chemistry in daily life

The Success Manual for General Chemistry Elizabeth Kean, Catherine Middlecamp, 1986

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Introduction to Organic and

Biological Chemistry Michael S. Matta, Antony C. Wilbraham, Dennis D. Staley, 1996 *Ebook: Biology*
 BROOKER, 2014-09-16 *Ebook Biology* *General Chemistry I as a Second Language* David R. Klein, 2005-03-16 Many students and instructors are overwhelmed by the vast amount of content and concepts presented in General Chemistry. Students often emerge from the course with little understanding of chemical concepts and must be retaught in subsequent courses. This supplemental text can be paired with Olmsted, Williams, Brady, Spencer or any other General Chemistry title. David Klein is a lecturer at Johns Hopkins University where he teaches Organic and General Chemistry. He is a dynamic and creative teacher and uses analogy to help students grasp difficult topics. Klein's unique informal voice and manner of presentation help students truly master key topics in this course. He is also the author of *Organic Chemistry as a Second Language*. Response to this book has been phenomenal. *Military Flight Aptitude Tests For Dummies* Terry J. Hawn, Peter Economy, 2013-06-04 The easy way to score high on the military aptitude flight test. The competition to become a military aviator is fierce. Candidates seeking entry into a military flight training program must first score well on a complicated service specific flight aptitude test. Now there's help. With practice exams and the most in depth instruction on the market, *Military Flight Aptitude Test For Dummies* gives future pilots, navigators, and aviation officers everything they need to score high and begin a career in military aviation. Plain English, in depth instruction, and test taking strategies for the various parts of each test. Practice exams for each of the service specific flight tests: AFOQT, SIFT, and ASTB. An overview of career options and paths to becoming an aviation officer. Whether you're looking to pursue an aviation career in the Air Force, Army, Navy, Marine Corps, or the Coast Guard, *Military Flight Aptitude Test For Dummies* has you covered. [Chemistry - The Central Science](#) James C. Hill, Bruce Edward Bursten, 2006 *Chemistry: The Central Science* is the most trusted book on the market. Its scientific accuracy, clarity, innovative pedagogy, functional problem solving, and visuals set this book apart. Brown, LeMay, and Bursten teach students the concepts and skills they need without overcomplicating the subject. A comprehensive media package that works in tandem with the text helps students practice and learn while providing instructors the tools they need to succeed. Publisher's description. *Study Guide for The Human Body in Health and Illness* Barbara Herlihy, 2013-11-27 Corresponding to the chapters in *The Human Body in Health and Illness*, 4th Edition by Barbara Herlihy, this study guide offers fun and practical exercises to help you review, understand, and remember basic A.P. Even if you find science intimidating, this book can help you succeed. Each chapter includes three parts: Mastering the Basics with matching, ordering, labeling, diagram reading, and coloring exercises; Putting It All Together including multiple choice quizzes and case studies; Challenge Yourself with critical thinking questions and puzzles. Textbook page references are included with the questions to make it easier to review difficult topics. Objectives at the beginning of each chapter reinforce the goals of the textbook and set a framework for study. UPDATED content matches the new and revised material in the 5th edition of the textbook. UPDATED coloring exercises improve your retention of the material. NEW exercises are included on the endocrine system.

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