

**Assessment 1****Chapter Test B****Chapter: Chemical Equations and Reactions**

**Part 1** In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. The precipitation of a slightly soluble solid compound from a solution of two ionic reactants is called the formation of a  
a. gas.  
b. precipitate.  
c. combustion reaction.  
d. halogen.
- \_\_\_\_\_ 2. To balance a chemical equation, it is permissible to adjust the  
a. coefficients.  
b. subscripts.  
c. formulae of the products.  
d. number of products.
- \_\_\_\_\_ 3. In a chemical equation, the symbol (aq) indicates that the substance is  
a. melted.  
b. dissolved in water.  
c. an acid.  
d. insoluble.
- \_\_\_\_\_ 4. The tendency for a replacement reaction to occur depends on the  
a. relative reactivities of the elements in the activity series themselves.  
b. temperature difference.  
c. valence electrons and mass.  
d. relative reactivities of the elements in the activity series themselves.
- \_\_\_\_\_ 5. The coefficients in a chemical equation  
a. indicate the number of moles of each substance that react.  
b. show the number of grams of each substance that react.  
c. are the molar masses of the substances.  
d. show the valence electrons for each atom.
- \_\_\_\_\_ 6. If metal X is more reactive than metal Y in the activity series, then  
a. X will replace ions of Y in solution.  
b. Y will replace ions of X in solution.  
c. Y will react with cold water.  
d. X will react with cold water.

# Guide Key Chemistry

**Daniel F McAuley**



## Guide Key Chemistry:

*Organic Chemistry Study Guide* Robert J. Ouellette, J. David Rawn, 2014-10-13 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned by doing It is difficult to learn by rote memorization and true understanding comes only from concentrated reading and working as many problems as possible In fact problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real world problems in the work place *How to Pass Advanced Higher Chemistry* Dr Sian Simmonds, 2021-03-08 Exam board SQA Level Advanced Higher Subject Chemistry First teaching August 2019 First exam Summer 2021 Trust Scotland's most popular revision guides to deliver the results you want The How to Pass series is chosen by students parents and teachers again and again This is the only study book that addresses the skills for Advanced Higher Chemistry as well as the knowledge b Recap and remember course content b Test your skills and knowledge b Practise exam style questions b Formal questions with mark allocations are provided at the end of each Key Area reflecting the types of questions you will face in the exam brbrb Get expert tips for exam success b Hints on how to achieve top marks and avoid mistakes are based on feedback in the SQA examiners Course Reports giving you insight into the marking process brbrb Teach yourself with confidence b Independent study has never been easier with clear explanations definitions of technical terms and answers to all questions at the back of the book br Jackson's compendium, a guide-book for the teaching of handwriting John Jackson (F.E.I.S.), 1895 **Guide to U. S. Government Publications**, 1980 **A Guide to the Literature of Chemistry** Evan Jay Crane, Austin McDowell Patterson, 1927 **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office, 1978 **The American Journal of Education**, 1863 *American Journal, and Annals of Education and Instruction*, 1863 *The Journal of the Kansas Medical Society* Kansas Medical Society, 1922 **An Evidence-based Guide to College and University Teaching** Aaron S. Richmond, Guy A. Boysen, Regan A. R. Gurung, 2016-06-10 What makes a good college teacher This book provides an evidence based answer to that question by presenting a set of model teaching characteristics that define what makes a good college teacher Based on six fundamental areas of teaching competency known as Model Teaching Characteristics outlined by The Society for the Teaching of Psychology STP this book describes how college faculty from all disciplines and at all levels of experience can

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