



Reaction Rates and Equilibrium Study Guide

Topics Covered:

Factors affecting reaction rates
Collision theory of reaction rates
Reaction progress diagrams (activation energy and transition states)
Rate laws and the method of initial rates
Chemical equilibrium
Equilibrium expressions and constants and their meaning
Product-favored vs. Reactant-favored
Shifting equilibrium and Le Chatelier's Principle
Quantitative calculations of equilibrium systems using ICE tables

Practice Questions: [Click here for an answer key](#)

- Describe how each of the following factors would affect the rate of "particle collisions" in your response.
 - Temperature decrease
 - Addition of iodide ions, I^- , as a catalyst
 - Increased reactant concentration
 - Crushing a solid into a powder (with larger surface area)
- The reaction progress diagram shows the energy changes that take place in the reaction of carbon monoxide gas with nitrogen dioxide gas.



Guide Reaction Rates And Equilibrium

Satyasree Gupta K



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Handbook of Biochemical Kinetics provides the underlying scaffolding of logic for kinetic approaches to distinguish rival models or mechanisms. The handbook also comments on techniques and their likely limitations and pitfalls as well as derivations of fundamental rate equations that characterize biochemical processes. Key Features: Over 750 pages devoted to theory and techniques for studying enzymic and metabolic processes; Over 1 500 definitions of kinetic and mechanistic terminology with key references; Practical advice on experimental design of kinetic experiments; Extended step by step methods for deriving rate equations; Over 1 000 enzymes complete with EC numbers, reactions catalyzed and references to reviews and or assay methods; Over 5 000 selected references to kinetic methods appearing in the Methods in Enzymology series; 72 page Wordfinder that allows the reader to search by keywords; Summaries of mechanistic studies on key enzymes and protein systems; Over 250 diagrams, figures, tables and structures. **NASA Technical Paper**, 1989

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and 4 Mike Smith,2015-11-30 Exam Board OCR Level AS A level Subject Chemistry First Teaching September 2015 First Exam Summer 2016 Written by experienced author Mike Smith this Student Guide for Chemistry Helps identify what you need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills Provides the content for generating individual revision notes

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