

**CHAPTER 7 REVIEW***Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. C In a Stock name such as iron(III) sulfate, the roman numeral tells us \_\_\_\_\_  
(a) how many atoms of Fe are in one formula unit  
(b) how many sulfate ions can be attached to the iron atom  
(c) the charge on each Fe ion  
(d) the total positive charge of the formula unit
2. C The result of changing a subscript in a correctly written chemical formula is to \_\_\_\_\_  
(a) change the number of moles represented by the formula  
(b) change the charges on the other ions in the compound  
(c) change the formula so that it no longer represents the compound it previously represented  
(d) have no effect on the formula
3. The explosive TNT has the molecular formula  $C_7H_5(NO_2)_3$ .  
4 elements a. How many elements make up this compound?  
6 oxygen atoms b. How many oxygen atoms are present in one molecule of  $C_7H_5(NO_2)_3$ ?  
21 atoms c. How many atoms in total are present in one molecule of  $C_7H_5(NO_2)_3$ ?  
 $4.2 \times 10^{24}$  atoms d. How many atoms are present in a sample of  $2 \times 10^{23}$  molecules of  $C_7H_5(NO_2)_3$ ?
4. How many atoms are present in each of these formula units?  
11 atoms a.  $Ca(HCO_3)_2$   
45 atoms b.  $C_{12}H_{22}O_{11}$   
10 atoms c.  $Fe(ClO_2)_3$   
9 atoms d.  $Fe(ClO_3)_2$
5.  $N_2O_5$  a. What is the formula for the compound dinitrogen pentoxide?  
carbon(IV) sulfide b. What is the Stock name for the covalent compound  $CS_2$ ?  
 $H_2SO_3$  c. What is the formula for sulfurous acid?  
phosphoric acid d. What is the name for the acid  $H_3PO_4$ ?

# Guided And Study Chemical Names And Formulas

**L Towner**



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## **Table of Contents Guided And Study Chemical Names And Formulas**

1. Understanding the eBook Guided And Study Chemical Names And Formulas
  - The Rise of Digital Reading Guided And Study Chemical Names And Formulas
  - Advantages of eBooks Over Traditional Books
2. Identifying Guided And Study Chemical Names And Formulas
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Guided And Study Chemical Names And Formulas
  - User-Friendly Interface
4. Exploring eBook Recommendations from Guided And Study Chemical Names And Formulas
  - Personalized Recommendations
  - Guided And Study Chemical Names And Formulas User Reviews and Ratings
  - Guided And Study Chemical Names And Formulas and Bestseller Lists

5. Accessing Guided And Study Chemical Names And Formulas Free and Paid eBooks
  - Guided And Study Chemical Names And Formulas Public Domain eBooks
  - Guided And Study Chemical Names And Formulas eBook Subscription Services
  - Guided And Study Chemical Names And Formulas Budget-Friendly Options
6. Navigating Guided And Study Chemical Names And Formulas eBook Formats
  - ePub, PDF, MOBI, and More
  - Guided And Study Chemical Names And Formulas Compatibility with Devices
  - Guided And Study Chemical Names And Formulas Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Guided And Study Chemical Names And Formulas
  - Highlighting and Note-Taking Guided And Study Chemical Names And Formulas
  - Interactive Elements Guided And Study Chemical Names And Formulas
8. Staying Engaged with Guided And Study Chemical Names And Formulas
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Guided And Study Chemical Names And Formulas
9. Balancing eBooks and Physical Books Guided And Study Chemical Names And Formulas
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Guided And Study Chemical Names And Formulas
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Guided And Study Chemical Names And Formulas
  - Setting Reading Goals Guided And Study Chemical Names And Formulas
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Guided And Study Chemical Names And Formulas
  - Fact-Checking eBook Content of Guided And Study Chemical Names And Formulas
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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