

BALANCING CHEMICAL EQUATIONS

Load the simulation *Balancing Chemical Equations*. This link is in your JupiterGrades messages <http://phet.colorado.edu/en/simulation/balancing-chemical-equations>

CRITICAL THINKING QUESTIONS

REVIEW:

MOLECULES: A GROUP OF ATOMS BONDED TOGETHER

ATOMS: THE BASIC UNIT OF A CHEMICAL ELEMENT

- Explore the *Balancing Chemical Equations* simulation. Discuss with your group what you find.
 - What are the tools you can use in the simulation to tell you when an equation is balanced?
Scale or bar graph

- For each balanced reaction, indicate the total number of molecules in the table below.

Reaction	Total Number of Molecules	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia	4	2
Separate Water	2	3
Combust Methane	3	3

- Is the number of total molecules on the left side of a balanced equation always equal to the number of total molecules on the right side of the equation? Explain your answer.
No because the atoms rearrange in a chemical reaction to form new molecules.

- For each balanced reaction, indicate the total number of atoms in the table below.

Reaction	Total Number of Atoms	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia	8	8
Separate Water	6	6
Combust Methane	9	9

- Is the number of total atoms on the left side of a balanced equation always equal to the number of total atoms on the right side of the equation?
Yes
- What is the same on the left and right side of a balanced equation? Explain your answer.
The numbers are all the same because they're balanced.
- As a group, play level 1 of the balancing equation game. Write down the strategies your group uses to balance chemical equations.

Guided Inquiry Balancing Chemical Equations

Jiyuan Zhang



Guided Inquiry Balancing Chemical Equations:

Decoding **Guided Inquiry Balancing Chemical Equations**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Guided Inquiry Balancing Chemical Equations**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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