

1.1^a. Concentration of a gas mixture.

A mixture of noble gases [helium (1), argon (2), krypton (3), and xenon (4)] is at a total pressure of 200 kPa and a temperature of 400 K. If the mixture has equal mole fractions of each of the gases, determine:

a) The composition of the mixture in terms of mass fractions.

Solution:

Basis: 100 kmole of the mixture

$$\begin{aligned}
 \text{ORIGIN} &:= 1 & \text{kmole} &:= 1000 \cdot \text{mole} \\
 \text{moles} &:= \begin{pmatrix} 25 \\ 25 \\ 25 \\ 25 \end{pmatrix} \cdot \text{kmole} & \text{MW} &:= \begin{pmatrix} 4 \\ 40 \\ 83.8 \\ 131.3 \end{pmatrix} \frac{\text{kg}}{\text{kmole}} & \text{mass} &:= \overrightarrow{(\text{moles} \cdot \text{MW})} \\
 \text{mass} &:= \begin{pmatrix} 100 \\ 1 \times 10^3 \\ 2.095 \times 10^3 \\ 3.283 \times 10^3 \end{pmatrix} \text{kg} & \text{total} &:= \sum_{i=1}^4 \text{mass}_i & \text{total} &:= 6.478 \times 10^3 \text{ kg} \\
 \text{mass_fraction} &:= \frac{\text{mass}}{\text{total}} & \text{mass_fraction} &:= \begin{pmatrix} 0.015 \\ 0.154 \\ 0.323 \\ 0.507 \end{pmatrix}
 \end{aligned}$$

b) The average molecular weight of the mixture.

Solution

$$\begin{aligned}
 \text{kmole} &:= 1000 \cdot \text{mole} \\
 M_{av} &:= \frac{\text{total}}{100 \cdot \text{kmole}} & M_{av} &= 0.065 \frac{\text{kg}}{\text{mol}}
 \end{aligned}$$

c) The total molar concentration.

Solution

$$\begin{aligned}
 P &:= 200 \cdot \text{kPa} & T &:= 400 \cdot \text{K} & \text{kPa} &:= 1000 \cdot \text{Pa} \\
 c &:= \frac{P}{R \cdot T} & R &:= 8.314 \cdot \text{kPa} \cdot \text{m}^3 \cdot \text{kmole}^{-1} \cdot \text{K}^{-1} \\
 c &:= 60.14 \frac{\text{mol}}{\text{m}^3}
 \end{aligned}$$

d) The mass density.

Solution

$$\begin{aligned}
 \rho &:= c \cdot M_{av} & \rho &= 3.896 \frac{\text{kg}}{\text{m}^3}
 \end{aligned}$$

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Water Environment Federation



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Solutions Manual to Accompany Mass-transfer Operations, Third Edition Robert Ewald Treybal, 1980

Handbook of Downstream Processing E. Goldberg, 2012-12-06 The last two decades have seen a phenomenal growth of the field of genetic or biochemical engineering and have witnessed the development and ultimately marketing of a variety of products typically through the manipulation and growth of different types of microorganisms followed by the recovery and purification of the associated products The engineers and biotechnologists who are involved in the full scale process design of such facilities must be familiar with the variety of unit operations and equipment and the applicable regulatory requirements This book describes current commercial practice and will be useful to those engineers working in this field in the design construction and operation of pharmaceutical and biotechnology plants It will be of help to the chemical or pharmaceutical engineer who is developing a plant design and who faces issues such as Should the process be batch or continuous or a combination of batch and continuous How should the optimum process design be developed Should one employ a new revolutionary separation which could be potentially difficult to validate or use accepted technology which involves less risk Should the process be run with ingredients formulated from water for injection deionized water or even filtered tap water Should any of the separations be run in cold rooms or in glycol jacketed lines to minimize microbial growth where sterilization is not possible Should the process equipment and lines be designed to be sterilized in place cleaned in place or should every piece be broken down cleaned and autoclaved after every turn **Catalog of Copyright Entries.**

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the costs of abatement justify the degree of abatement achieved This book is one of the volumes of the Handbook of Environmental Engineering series The principal intention of this series is to help readers formulate answers to the last two questions above The traditional approach of applying tried and true solutions to specific pollution problems has been a major contributing factor to the success of environmental engineering and has accounted in large measure for the establishment of a methodology of pollution control However the realization of the ever increasing complexity and interrelated nature of current environmental problems renders it imperative that intelligent planning of pollution abatement systems be undertaken

Design of Water Resource Recovery Facilities, Manual of Practice No.8, Sixth Edition Water Environment Federation, 2017-09-29 Complete Coverage of the State of the Art in Water Resource Recovery Facility Design Featuring contributions from hundreds of wastewater engineering experts this fully updated guide presents the latest in facility planning configuration and design Design of Water Resource Recovery Facilities WEF Manual of Practice No 8 and ASCE Manuals and Reports on Engineering Practice No 76 Sixth Edition covers key technical advances in wastewater treatment including Advances with membrane bioreactors applications Advancements within integrated fixed film activated sludge IFAS systems and moving bed biological reactors systems Biotrickling filtration for odor control Increased use of ballasted flocculation Enhanced nutrient control systems Sidestream nutrient removal to reduce the loading on the main nutrient removal process Use and application of wireless instrumentation Use and application of modeling wastewater treatment processes for the basis of design and evaluations of alternatives Process design and disinfection practices to minimize generation of TTHMs and other organics monitored for potable water quality Approaches to minimizing biosolids production and advances in biosolids handling including effective thermal hydrolysis and improvements in sludge thickening and dewatering technologies Increasing goals toward energy neutrality and driving net zero Trend toward resource recovery

Handbook of Separation Process Technology Ronald W. Rousseau, 1987-05-13 Surveys the selection design and operation of most of the industrially important separation processes Discusses the underlying principles on which the processes are based and provides illustrative examples of the use of the processes in a modern context Features thorough treatment of newer separation processes based on membranes adsorption chromatography ion exchange and chemical complexation Includes a review of historically important separation processes such as distillation absorption extraction leaching and crystallization and considers these techniques in light of recent developments affecting them *Manual on Indoor Air Quality* R. C. Diamond, 1984

Computer Programs for Chemical Engineering Education: R. Gordon. Transport, 1972 **Chemical Engineering Design** Ray Sinnott, Gavin Towler, 2009-05-15 Chemical Engineering Design is one of the best known and most widely adopted texts available for students of chemical engineering It completely covers the standard chemical engineering final year design course and is widely used as a graduate text The hallmarks of this renowned book have always been its scope practical emphasis and closeness to the curriculum That it is written by practicing chemical

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Encyclopedia of Energy Technology and the Environment Attilio Bisio, Sharon Boots, 1995

Proceedings of the Estonian Academy of Sciences, Chemistry, 2004-09

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