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## Mastering Biology Ch 04 HW

Most organic compounds contain carbon and \_\_\_\_\_. - answer Hydrogen

What is the molar ratio of methionine to glycine in the samples? - answer  $1.8 \times 10^{-3}$  mole of methionine per mole of glycine

How many molecules of glycine are present in 1.0 mole of glycine? - answer  $6.02 \times 10^{23}$  molecules

For every 1.0 mole of glycine in the sample, how many molecules of methionine are present? - answer  $1.08 \times 10^{21}$  molecules of methionine

Which amino acid is present in higher amounts than glycine in the samples? - answer Alanine

If there is 1.0 mole of glycine present in the sample, how many more molecules of alanine are present? - answer  $6.02 \times 10^{22}$

If one mole each of  $\text{CH}_4$ ,  $\text{NH}_3$ ,  $\text{H}_2\text{S}$ , and  $\text{CO}_2$  is added to 1 liter of water in a flask (1 liter of water = 55.5 moles of  $\text{H}_2\text{O}$ ), how many moles of hydrogen, carbon, oxygen, nitrogen, and sulfur are in the flask? Round your answers to the nearest whole number. - answer Hydrogen-120

Carbon-2

Oxygen-58

Nitrogen-1

Sulfur-1

Glycine's molecular formula is  $\text{C}_2\text{H}_5\text{NO}_2$ . How many moles of each element would be needed to make 1.0 mole of glycine? - answer 2 moles carbon, 5 moles hydrogen, 1 mole nitrogen, and 2 moles oxygen

What is the maximum number of moles of glycine ( $\text{C}_2\text{H}_5\text{NO}_2$ ) that could be made in the flask if no other molecules were made? - answer 1 mole of glycine

If only serine ( $\text{C}_3\text{H}_7\text{NO}_3$ ) was made, which element would be used up first in the flask? - answer Carbon

What is the maximum number of moles of serine that could be made in the flask if no other molecules were made? - answer  $2/3$  (0.67) of a mole of serine

If only methionine ( $\text{C}_5\text{H}_{11}\text{NO}_2\text{S}$ ) was made, which element would be used up first? - answer Carbon

# Mastering Biology Ch 6 Answers

**S. Chand Experts**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the 'S. Chand Experts' text.

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