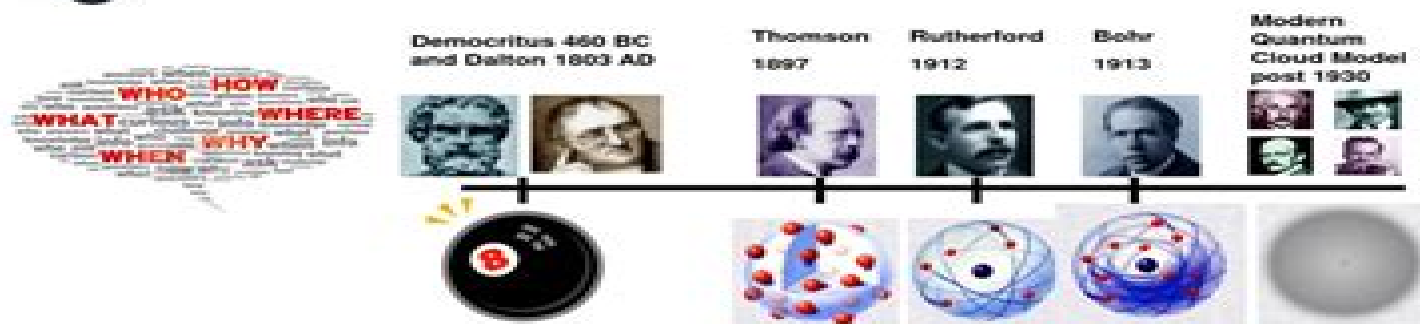




History of the Atom Timeline



John Dalton (1766 – 1844):

John Dalton was an English chemist. His ideas form the atomic theory of matter. Here are his ideas.

- All elements are composed (made up) of atoms. It is impossible to divide or destroy an atom.
- All atoms of the same elements are alike. (One atom of oxygen is like another atom of oxygen.)
- Atoms of different elements are different. (An atom of oxygen is different from an atom of hydrogen.)
- Atoms of different elements combine to form a compound. These atoms have to be in definite whole number ratios. For example, water is a compound made up of 2 atoms of hydrogen and 1 atom of oxygen (a ratio of 2:1). Three atoms of hydrogen and 2 atoms of oxygen cannot combine to make water.

Dalton's Atomic Theory



1. What is the name of John Dalton's theory? **THE ATOMIC THEORY OF MATTER**
2. What are elements made of? **ELEMENTS ARE MADE OF ATOMS**
3. An atom of hydrogen and an atom of carbon are **DIFFERENT**
4. What are compounds made of? **COMPOUNDS ARE MADE OF ATOMS OF TWO OR MORE DIFFERENT ELEMENTS**
5. The ratio of atoms in HCl is: a) 1:3 b) 2:1 c) **1:1**
6. The ratio of atoms in H₂O is: a) 1:3 b) **2:1** c) 1:1 **2 atoms of hydrogen and 1 atom of oxygen**

History Of The Atom Answer Key

Guillaume Favre



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Bernard Pullman, 2001 The concept of the atom is very close to scientific bedrock the deepest and most fundamental fact about the nature of reality This book presents the whole panorama of the atomic hypothesis and its place in Western civilization from its origins in early Greek philosophy 2500 years ago to the definitive proof through direct microscopic imaging of since atoms about ten years ago

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