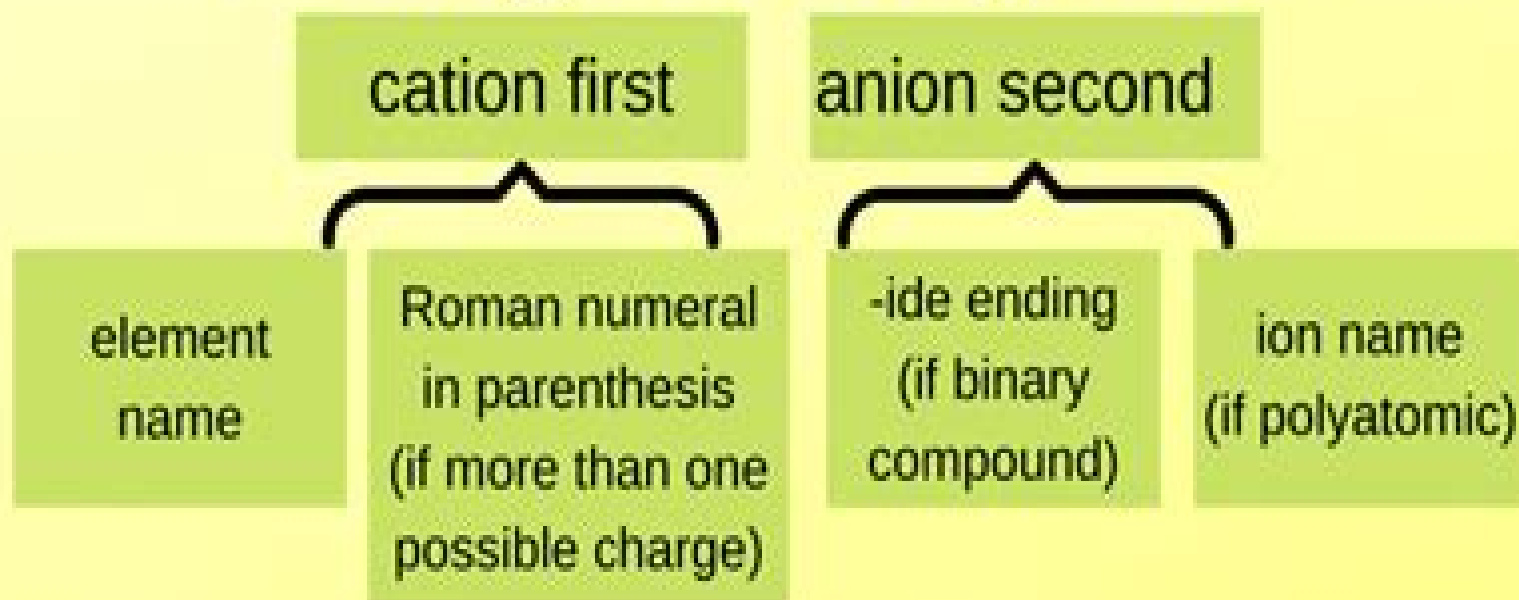


Naming Ionic Compounds



strontium chloride



cobalt(II) chloride



platinum(II) arsenate

Holt Chemistry Ions And Ionic Compounds

**David W. Oxtoby, H. Pat Gillis, Laurie J.
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From Classical to modern chemistry Peter John Turnbull Morris, 2002

Lab Experiments Modern Chemistry Harold Clark Metcalfe, 1982

Fundamentals of Chemistry: A Modern Introduction (1966) Frank Brescia, 2012-12-02 Fundamentals of Chemistry A Modern Introduction focuses on the formulas processes and methodologies used in the study of chemistry The book first looks at general and historical remarks definitions of chemical terms and the classification of matter and states of aggregation The text then discusses gases Ideal gases pressure of a gas confined by a liquid Avogadro's Law and Graham's Law are described The book also discusses aggregated states of matter atoms and molecules chemical equations and arithmetic thermochemistry and chemical periodicity The text also highlights the electronic structures of atoms Quantization of electricity spectra of elements quantization of the energy of an electron associated with nucleus the Rutherford Bohr nuclear theory hydrogen atom and representation of the shapes of atomic orbitals are explained The text also highlights the types of chemical bonds hydrocarbons and their derivatives intermolecular forces solutions and chemical equilibrium The book focuses as well on ionic solutions galvanic cells and acids and bases It also discusses the structure and basicity of hydrides and oxides The reactivity of hydrides charge of dispersal and basicity effect of anionic charge inductive effect and basicity and preparation of acids are described The book is a good source of information for readers wanting to study chemistry

Introduction to Modern Chemistry Edward Florian Neuzil, 1968

Modern Chemistry John F. Baxter, Luke Eby Steiner, 1959

Elementary Modern Chemistry Wilhelm Ostwald, Harry

Wheeler Morse, 1909

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