OMPTER (2) STUDY GUIDE FOR CONTENT MASTERS

## Section 12.3 Limiting Reactants

in your textbook, read about any mactions stop and how to determine the Uniting reaction.

Study the diagram showing a chemical reaction and the elemical equation that represents the traction. Then complete the table. Show your calculations for questions 25-27 in the space below the table.

$$\Box + 00 + 0 \rightarrow 0 + 00$$
  
 $0,+200 \rightarrow 200$ 

The media masses of O<sub>2</sub>, NO, and NO<sub>2</sub> are 32.00 g/mel, 30.01 g/mel, and 46.01 g/mel, emporatedly.

Amount of O <sub>2</sub>	Annual of NO	Amount of Mily	Similing Number	Amount and Name of Quees Resident
1 molecule	Teolesies	2 molecules	mine	none
Challetales	4 mileciles	4 molecules	NO	2 milesia 0 <sub>y</sub>
I molecules	1 militaire	1. 4 molecules	2. 02	1, 4 molecules NO
3.88 mol -	2.00 mail	4. 2.00 mol	5. none	6. none
438 mol	450 mail	7, 4.00 mail	E. NO	9, 2.00 mol 0 <sub>2</sub>
3.00 mail	7.00 mult	16, 7.00 mol	11, NO	12, 1.50 mail 0 <sub>2</sub>
1.00 mol	400 mil	13. 2.00 mol	14. O <sub>3</sub>	15. 2.00 mol NO
1.500 mail	0.200 mail	16, 0,200 mol	17, 100	18. 0.400 mol O <sub>2</sub>
name	mag	19. 92.02 g	20. none	21. none
26.00 (6	mong	22, 46.01 g	20. 02	34, 50.12 g NO
mm <sub>1</sub>	20.00 g	25, 28.74 g	26. 02	37, 1,34 g NO

balanced equation mole ratio = 2 mol NO/1 mol O2

10.00 g.O<sub>2</sub> × 1 mol O<sub>2</sub>/32.00 g.O<sub>2</sub> = 0.3125 mol O<sub>2</sub>

20.00 p.NO × 1 mai NO/30.01 p.NO = 0.6664 mai NO

actual mole ratio = 0.6664 mol NO/0.3125 mol O<sub>2</sub> = 2.132 mol NO/1.000 mol O<sub>2</sub> Secause the actual mole ratio of NO:O<sub>2</sub> is larger than the balanced equation mole ratio of NO:O<sub>3</sub>, there is an excess of NO; O<sub>3</sub> is the limiting reactant.

Mass of NO used = 0.3125 gpai-0; × 2 mol NO<sup>1</sup>1 gpai-0; = 0.6250 mol NO 0.6250 gpai-NO × 30.01 g NO<sup>1</sup>1 gpai-400 = 18.76 g NO

Mass of NO<sub>2</sub> produced = 0.6250 gpsi.WO<sub>2</sub> × 46.01 g NO<sub>2</sub>/1 gpsi.WO<sub>2</sub> = 28.76 g NO<sub>2</sub> Excess NO = 20.00 g NO - 18.76 g NO = 1.24 g NO

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Chemistry Matter and Change + Chapter 12 ... 29.



#### Section 12.4 Percent Yield

In your textlisiol, resel about the pields of products.

Study the diagram and the example problem.

mass of product from experimental measurement

mais of product predicted from assistionietric calculation using

- at mass of reactions.
- 5. 4 step main to main conversion
  - 1. Sems the belanced chemical equation.
    - Calculate the number of moles of reactions, using moles mass.
    - Calculate the number of moles of product, using the appropriate mole ratio.
    - Calculate the mass of product, using the reciprocal of molar mass.

Example Problem: The following chemical equation represents the production of gallium oriole, a substance used in the manufacturing of some semiconductor devices.

In one experiment, the reaction yielded 7.42 g of the enals from a 7.85 g sample of gallium. Describes the percent yield of this reaction. The molar masses of Ga and Ga<sub>2</sub>O<sub>2</sub> are 46.72 glossl and 167.46 glossl, respectively.

Use the information in the diagram and example problem to evaluate each value or expression below. If the value or expression is correct, write correct. If it is incorrect, write the correct value or expression.

- 1. actual right unknown 7.42 g Ga<sub>2</sub>O<sub>3</sub>
- 2. mass of touctast: Total y the correct
- matter of motion of reactant: 7.00 g/G<sub>0</sub> × min(T<sub>2</sub> g/G<sub>0</sub>).
   mod G<sub>0</sub>.
   mod G<sub>0</sub>.
   mod G<sub>0</sub>.
   mod G<sub>0</sub>.
- 4. number of moles of product 0.000 mol  $G_0 \times \frac{2 \bmod G_0/O_0}{1 \bmod G_0} = \frac{0.100 \bmod G_0 \times 2 \bmod G_0}{2 \bmod G_0/O_0/4 \bmod G_0}$
- 5. theoretical yield. 0:0500 and  $O_{m_0}O_{\gamma} \times \frac{107.44 \pm O_{m_0}O_{\gamma}}{1 \text{ and } O_{m_0}O_{\gamma}}$  connect.
- percent yield: 41.7 g Ga<sub>2</sub>O<sub>3</sub> × 100
   7.42 g Ga<sub>2</sub>O<sub>3</sub>/9.37 g Ga<sub>2</sub>O<sub>3</sub> × 100
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# **Limiting Reactants Study Guide**

**Kathleen Armour** 

## **Limiting Reactants Study Guide:**

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carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students The Complete Chemistry Study Guide and Note Cards and MCAT Konstantinos Papadopoulos, 2012-07-06 ASAP Chemistry: A Quick-Review Study Guide for the AP **Exam** The Princeton Review, 2019-02-12 Looking for sample exams practice questions and test taking strategies Check out our extended in depth AP chem prep guide Cracking the AP Chemistry Exam LIKE CLASS NOTES ONLY BETTER The Princeton Review's ASAP Chemistry is designed to help you zero in on just the information you need to know to successfully grapple with the AP test No questions no drills just review Advanced Placement exams require students to have a firm grasp of content you can t bluff or even logic your way to a 5 Like a set of class notes borrowed from the smartest student in your grade this book gives you exactly that No tricks or crazy stratagems no sample essays or practice sets Just the facts presented with lots of helpful visuals Inside ASAP Chemistry you ll find Essential concepts terms and functions for AP Chem all explained clearly concisely Diagrams charts and graphs for quick visual reference A three pass icon system designed to help you prioritize learning what you MUST SHOULD and COULD know in the time you have available Ask Yourself questions to help identify areas where you might need extra attention A resource that s perfect for last minute exam prep and for daily class work Topics covered in ASAP Chemistry include Atomic structure Covalent bonding intermolecular forces Thermochemistry Acids bases and more Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl Paul B. Kelter, 1993 "O" Level Study Guide - Chemistry Quite Easily Done, Organic Chemistry Reactions: A Study Guide Cybellium, 2024-10-26 Designed for professionals students and enthusiasts alike our comprehensive books empower you to stay ahead in a rapidly evolving digital world Expert Insights Our books provide deep actionable insights that bridge the gap between theory and practical application Up to Date Content Stay current with the latest advancements trends and best practices in IT Al Cybersecurity Business Economics and Science Each guide is regularly updated to reflect the newest developments and challenges Comprehensive Coverage Whether you re a beginner or an advanced learner Cybellium books

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Study Guide for Introductory Chemistry, a Foundation, Introductory Chemistry, Basic Chemistry Iris Stovall, Steven S. Zumdahl, 1996

Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through Limiting Reactants Study Guide

In some sort of inundated with monitors and the cacophony of fast interaction, the profound energy and psychological resonance of verbal artistry often diminish in to obscurity, eclipsed by the constant onslaught of sound and distractions. Yet, set within the lyrical pages of **Limiting Reactants Study Guide**, a interesting perform of fictional brilliance that impulses with organic feelings, lies an unforgettable trip waiting to be embarked upon. Published by way of a virtuoso wordsmith, this interesting opus books visitors on an emotional odyssey, gently exposing the latent potential and profound affect stuck within the complex web of language. Within the heart-wrenching expanse with this evocative evaluation, we will embark upon an introspective exploration of the book is central styles, dissect its interesting writing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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