

Chemistry Study Guide

Periodic Table of the Elements

The periodic table displays elements organized by atomic number, chemical properties, and electron configuration. It includes the main groups, transition metals, and the lanthanide and actinide series at the bottom.

Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎📌

Matter:

Anything that has **mass** and **occupies space** ✎🌐

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

Lavc Chemistry Study Guide

Jin-Ying Zhang



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